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All I common descriptions			

# Student Exploration: Ionic Bonds

Directions: Follow the instructions to go through the simulation. Respond to the questions and prompts in the orange boxes.

**Vocabulary:** chemical family, electron affinity, ion, ionic bond, metal, nonmetal, octet rule, shell, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

 Nate and Clara are drawing pictures with markers. There are 8 markers in a set. Nate has 9 markers and Clara has 7. What can Nate and Clara do so that each of them has a full set?

Give one marker to Clara

Maggie is sitting at a table with Fred and Florence. Maggie has 10 markers, but Fred and Florence each have only 7 markers. How can they share markers so each has 8?

Maggie can give 1 marker to Fred, and one marker to Florence.

#### Gizmo Warm-up

Just like students sharing markers, atoms sometimes share or swap electrons. By doing this, atoms form bonds. The *lonic Bonds* Gizmo allows you to explore how **ionic bonds** form.

To begin, check that Sodium (Na) and Chlorine (CI) are selected from the menus at right. Click Play ( E) to see electrons orbiting the nucleus of each atom. (Note: These atom models are simplified and not meant to be realistic.)



Each atom consists of a central nucleus and several shells that contain electrons. The outermost electrons
are called valence electrons. (Inner electrons are not shown.)

How many valence electrons does each atom have?

Sodium:

Chlorine:

7

- Click Pause ( !L.). Elements can be classified as metals and nonmetals. Metals do not hold on to their valence electrons very tightly, while nonmetals hold their electrons tightly. Electron affinity is a measure of how tightly the valence electrons are held.
  - A. Try pulling an electron away from each atom. Based on this experiment, which atom is a metal?

Sodium

Which is a nonmetal?

Chlorine

B. Try moving an electron from the metal to the nonmetal. What happens?

Sodium's electron is transferred to Chlorine's outer ring of electrons.

## **Gizmos Answers Ionic Bonds**

**JA Banks** 

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#### **Table of Contents Gizmos Answers Ionic Bonds**

- 1. Understanding the eBook Gizmos Answers Ionic Bonds
  - The Rise of Digital Reading Gizmos Answers Ionic Bonds
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Gizmos Answers Ionic Bonds
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gizmos Answers Ionic Bonds
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Gizmos Answers Ionic Bonds

- Personalized Recommendations
- Gizmos Answers Ionic Bonds User Reviews and Ratings
- Gizmos Answers Ionic Bonds and Bestseller Lists
- 5. Accessing Gizmos Answers Ionic Bonds Free and Paid eBooks
  - Gizmos Answers Ionic Bonds Public Domain eBooks
  - Gizmos Answers Ionic Bonds eBook Subscription Services
  - Gizmos Answers Ionic Bonds Budget-Friendly Options
- 6. Navigating Gizmos Answers Ionic Bonds eBook Formats
  - o ePub, PDF, MOBI, and More
  - Gizmos Answers Ionic Bonds Compatibility with Devices
  - Gizmos Answers Ionic Bonds Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gizmos Answers Ionic Bonds
  - Highlighting and Note-Taking Gizmos Answers Ionic Bonds
  - Interactive Elements Gizmos Answers Ionic Bonds
- 8. Staying Engaged with Gizmos Answers Ionic Bonds
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gizmos Answers Ionic Bonds
- 9. Balancing eBooks and Physical Books Gizmos Answers Ionic Bonds
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gizmos Answers Ionic Bonds
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Gizmos Answers Ionic Bonds
  - Setting Reading Goals Gizmos Answers Ionic Bonds
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gizmos Answers Ionic Bonds

- Fact-Checking eBook Content of Gizmos Answers Ionic Bonds
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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