

Name: _____ Period: _____

Properties of Water Worksheet

Word Bank:

covalent	solvent	deposition	dissolve	cohesion	surface	cooler	universal
tension	polar	viscosity	condensation	negatively	adhesion	positively	sublimation

1. The hydrogen and oxygen atoms are held together by covalent bonds.
2. The electrons are not shared equally creating a polar molecule.
3. The polarity of water allows it to dissolve most substances. Because of this it is referred to as the universal solvent.
4. Water molecules stick to other water molecules. This property is called cohesion.
5. Hydrogen bonds form between adjacent water molecules because the positively charged hydrogen end of one water molecule attracts the negatively charged oxygen end of another water molecule.
6. Water molecules stick to other materials due to its polar nature. This property is called adhesion.
7. surface tension creates the skin-like surface formed due to the polar nature of water.
8. viscosity is the tendency for fluids to resist flow.
cooler water is more viscous than warmer water, allowing floating organisms to use less energy to keep from sinking.
9. condensation is when water changes from a gas to a liquid.
10. sublimation is when water changes from a solid directly to a gas.
11. deposition is when water changes from a gas directly to a solid.
12. Why does ice float?

water is less dense in solid form because hydrogen bonding between water molecules pushed molecules further apart into a set crystalline pattern

13. List the properties that ocean water has due to the fact that it contains salt.

higher density
conductivity
decreased surface tension

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Eriola Betiku, Mofoluwake M. Ishola



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Table of Contents 171 Liquid Water And Its Properties Section Review Answers

1. Understanding the eBook 171 Liquid Water And Its Properties Section Review Answers
 - The Rise of Digital Reading 171 Liquid Water And Its Properties Section Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 171 Liquid Water And Its Properties Section Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 171 Liquid Water And Its Properties Section Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 171 Liquid Water And Its Properties Section Review Answers
 - Personalized Recommendations

- 171 Liquid Water And Its Properties Section Review Answers User Reviews and Ratings
- 171 Liquid Water And Its Properties Section Review Answers and Bestseller Lists
- 5. Accessing 171 Liquid Water And Its Properties Section Review Answers Free and Paid eBooks
 - 171 Liquid Water And Its Properties Section Review Answers Public Domain eBooks
 - 171 Liquid Water And Its Properties Section Review Answers eBook Subscription Services
 - 171 Liquid Water And Its Properties Section Review Answers Budget-Friendly Options
- 6. Navigating 171 Liquid Water And Its Properties Section Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - 171 Liquid Water And Its Properties Section Review Answers Compatibility with Devices
 - 171 Liquid Water And Its Properties Section Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 171 Liquid Water And Its Properties Section Review Answers
 - Highlighting and Note-Taking 171 Liquid Water And Its Properties Section Review Answers
 - Interactive Elements 171 Liquid Water And Its Properties Section Review Answers
- 8. Staying Engaged with 171 Liquid Water And Its Properties Section Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 171 Liquid Water And Its Properties Section Review Answers
- 9. Balancing eBooks and Physical Books 171 Liquid Water And Its Properties Section Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 171 Liquid Water And Its Properties Section Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 171 Liquid Water And Its Properties Section Review Answers
 - Setting Reading Goals 171 Liquid Water And Its Properties Section Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 171 Liquid Water And Its Properties Section Review Answers
 - Fact-Checking eBook Content of 171 Liquid Water And Its Properties Section Review Answers

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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