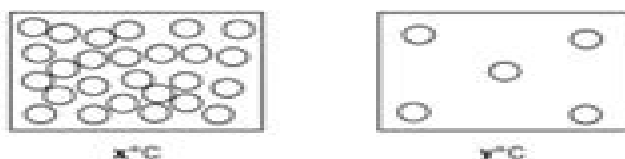


9. The diagrams below show the particles of hexane at atmospheric pressure but at two different temperatures. Hexane melts at -95°C and boils at 70°C .



What could be the values of x and y ?

	x	y
A	-100	0
B	-50	10
C	0	100
D	10	-50

(C)

Section B: Structured Questions

Answer the following questions in the spaces provided.

1. The interconversions between the three states of matter are represented by the diagram below.



- (a) Name the process in

(i) X : Boiling/ Evaporation (ii) Y : Freezing (iii) Z : Condensation

- (b) In which of the change(s) of state (X , Y and/or Z) is energy being given out?

Y and Z

2. In terms of the kinetic particle theory, explain the following.

- (a) A liquid can take the shape of its container.

The particles in a liquid are not held in fixed positions and can move freely throughout the liquid.

- (b) A tub of ice is heavier than a tub filled with water vapour.

The particles of ice, a solid, are very closely packed together. The particles of water vapour, a gas, are spread far apart from one another. Thus, ice has a higher density than water vapour of the same volume.

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