

Name: _____

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S'mores Stoichiometry Lab

Introduction:

In this experiment, you will explore the principles of stoichiometry by building S'mores, the delicious, chocolate, marshmallow, and graham cracker treats.



Definitions: 'Stoichio' means element and 'metry' means the process of measuring. The mass and quantity relationships among reactants and products in a reaction are found using the process of stoichiometry.

Purpose:

To determine the limiting reactant in the synthesis of S'mores.

Prelab Questions:

1. What does a chemical equation tell us?
2. What do the coefficients in a balanced chemical equation represent?

Materials:

Miniature Chocolate bars
Marshmallows
Bunsen Burner

Graham cracker Square
Paper Towels
Wooden skewers

Reference Data:

| Substance | Symbol | Unit mass | Package Mass |
|-------------------------|--------|-----------|--------------|
| Graham Cracker Squares | Gr | 8.50 g | 408 g |
| Marshmallow | Mm | 7.125 g | 285 g |
| Miniature Chocolate Bar | Ch | 12.00 g | 348 g |

** for the purposes of this lab the UNIT mass and MOLAR mass are the same thing

Procedure:

1. Perform a synthesis reaction (forming ~~one~~ s'more). For this lab, we will consider a s'more to consist of 2 graham cracker squares, 1 marshmallow and 1 miniature chocolate bar.
 - a) Determine the formula for a s'more
 - b) write the balanced chemical equation using the symbols above
 - c) According to the balanced equation complete the following ratios:

| | |
|---------------------|-------------------------|
| _____ Gr = _____ Mm | _____ Gr = _____ S'more |
| _____ Ch = _____ Mm | _____ Mm = _____ S'more |
| _____ Ch = _____ Gr | _____ Ch = _____ S'more |
2. Calculating the unit mass of the s'more.
 - a) Calculate the mass on ONE s'more.
 - b) Is there a relationship between the mass of a S'more and the masses of the reactants used to make it?
 - c) If so, what is the relationship?
 - d) What law have you studied in this course that might define this relationship?
3. Predicting the limiting reactant:
 - a) Complete the data table below based on the materials provided for your group.

| Substance | Quantity received | # of S'mores it COULD make |
|-------------------------|-------------------|----------------------------|
| Graham Cracker Squares | | |
| Marshmallow | | |
| Miniature Chocolate Bar | | |
 - b) What will the limiting reactant be?
 - c) How many of the other substances will remain
 - d) What are the masses of each excess reactant that will remain once the reaction is complete?

Apply Stoichiometry Lab Answer Key

**Tatsuki Ohji, Tohru Sekino, Koichi
Niihara**



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in thinking through what reform in undergraduate science teaching might look like and how we help faculty to implement such reform

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