

Solving Proportions Worksheet

Name: _____

Date: _____ Section: _____

Each problem must to be set up this way:

1. Write the proportion.

$$\frac{8}{3} = \frac{192}{n}$$

2. Write the cross products

$$8 * n = 192 * 3$$

3. Multiply

$$8n = 576$$

4. Undo multiplication by using division

$$8n = 576$$

$$\div 8 \quad \div 8$$

5. Divide

$$n = 72$$

Solve each proportion. Be sure to set it up the correct way and show all work.

1. $\frac{4}{9} = \frac{10}{x}$

2. $\frac{5}{2} = \frac{6}{x}$

3. $\frac{5}{2} = \frac{2}{x}$

4. $\frac{21}{27} = \frac{x}{18}$

5. $\frac{15}{21} = \frac{20}{y}$

6. $\frac{26}{b} = \frac{39}{9}$

7. $\frac{h}{108} = \frac{7}{18}$

8. $\frac{45}{792} = \frac{70}{w}$

9. $\frac{16}{120} = \frac{j}{15}$

10. $\frac{350}{p} = \frac{1050}{60}$

11. $\frac{g}{1134} = \frac{27}{729}$

12. $\frac{40}{65} = \frac{z}{104}$

Practice 8 1 Ratios And Proportions Answers

Henry Evers



Practice 8 1 Ratios And Proportions Answers:

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web trimyristin compound from nutmeg seed trimyristin compounds contained in nutmeg seed is very important to be isolated because it has various benefits trimyristin along with myristic acid myristicin and elemicin have antioxidant anticonvulsant analgesic anti inflammatory antidiabetic antibacterial and anti fungal activity

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid

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web purpose the purpose of this lab is to extract trimyristin from nutmeg as well as synthesizing myristic acid from the extracted trimyristin using base hydrolysis and naoh acidification using hcl is necessary to characterize the product by finding the melting point

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web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

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