

REGULAR

REGULAR STUDENT FEES AND OTHER CHARGES

EFFECTIVE 2021/2022 ACADEMIC YEAR

	ITEM	YEAR 1			YEAR 2			YEAR 3			YEAR 4
		SEM 1	SEM 2	TOTAL PA	SEM 1	SEM 2	TOTAL PA	SEM 1	SEM 2	TOTAL PA	(6 MONTHS)
		Amount (Ksh.)			Amount (Ksh.)			Amount (Ksh.)			
1.	Registration Fee	2,000.00	-	2,000.00	-	-	-	-	-	-	-
2.	Activity Fee	1,000.00	1,000.00	2,000.00	1,000.00	1,000.00	2,000.00	1,000.00	1,000.00	2,000.00	5,000.00
3.	Examination Fee	5,000.00	5,000.00	10,000.00	5,000.00	5,000.00	10,000.00	5,000.00	5,000.00	10,000.00	2,000.00
4.	Library and ICT services	2,000.00	2,000.00	4,000.00	2,000.00	2,000.00	4,000.00	2,000.00	2,000.00	4,000.00	2,000.00
5.	Tuition fee	15,000.00	15,000.00	30,000.00	15,000.00	15,000.00	30,000.00	15,000.00	15,000.00	30,000.00	15,000.00
6.	Research and Development Fee	5,000.00	5,000.00	10,000.00	5,000.00	5,000.00	10,000.00	5,000.00	5,000.00	10,000.00	5,000.00
7.	Transport Fee	6,000.00	6,000.00	12,000.00	6,000.00	6,000.00	12,000.00	6,000.00	6,000.00	12,000.00	3,000.00
8.	Maintenance Fee	5,000.00	5,000.00	10,000.00	5,000.00	5,000.00	10,000.00	5,000.00	5,000.00	10,000.00	5,000.00
9.	Student Identity Card	700.00	-	700.00	-	-	-	-	-	-	-
	TOTAL KSHS.	41,700.00	39,000.00	80,700.00	39,000.00	39,000.00	78,000.00	39,000.00	39,000.00	78,000.00	39,000.00

Charges for meals and accommodation shall be payable directly at the respective Campus at applicable rates. The respective College bank account details shall be provided upon admission for this purpose.

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Roman Wölfel



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web 1 define each remember oil rig oxidation is loss of e reduction is gain of e a oxidation b reduction c oxidizing agent d reducing agent 2 write half reactions for each of the following atoms or ions label each as oxidation or reduction use the activity series chart a al b ba2 c br 2 d ca e ga3 f h 2 g h

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web during oxidation and reduction the oxidation numbers of atoms change if an atom is oxidized its oxidation number increases ie it becomes more ve or less ve if an atom is reduced its oxidation number decreases ie it becomes less ve or more ve

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web write the oxidation and reduction reactions for each redox reaction the first one is done for you 23 fe2 co co2 fe oxidation co co2 2e reduction fe2 2e fe 24 3 ag ni ni3 3 ag oxidation reduction 25 cu2 pb pb2 cu oxidation reduction 26 o 2

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent
 $\text{I}_2\text{O}_5 + 5\text{CO} \rightarrow \text{I}_2 + 5\text{CO}_2$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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 $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$ oxidized reducing agent O_2 to O^{2-} reduced ox ag $2\text{Li} + \text{S} \rightarrow 2\text{Li}_2\text{S}$ oxidized red ag S^{0} to S^{2-} reduced ox ag $3\text{C} + 2\text{CS}_2 \rightarrow 3\text{C}_2\text{S}_2$

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