

THE CHEMISTRY OF LIFE

Living matter is composed of many different elements that are made up of small units called atoms. An atom of any element has a nucleus that is surrounded by particles called electrons. Each electron has a negative charge. A nucleus usually has two types of particles, protons and neutrons. Each proton has a positive charge. Neutrons have no charge. The number of protons in a nucleus of atom is called its atomic number. When you add up the number of protons and neutrons together, you get what is called the atomic mass. The chart below shows six elements that make up most of your body.

Use this information about atoms and the facts in the chart to answer the questions that follow the chart.

Element	Element Symbol	Atomic number	Atomic mass	Percent in the human body
Hydrogen	H	1	1	9.5
Carbon	C	6	12	18.5
Nitrogen	N	7	14	3.0
Oxygen	O	8	16	65.0
Phosphorus	P	15	31	1
Sulfur	S	16	32	0.3

1. What element is the most abundant in the human body? What is its symbol?

2. What is the atomic number of the element in question 1?

3. What makes up to 18.5 percent of the body? What is its symbol?

4. Write the atomic number of phosphorous. How many protons does it have?

5. Finish the word equation

6. Atomic mass = the number of _____ + the number of _____
7. What is the atomic mass of nitrogen?

8. How many neutrons are there in the nucleus of a nitrogen atom?

9. Write the symbol of the element found in only 0.3 percent of human body.

10. Which element in the chart has no neutrons?

Chemistry Of Life Short Answer

Dr. P.S. Verma & Dr. B.P. Pandey



Chemistry Of Life Short Answer:

Chemistry in Daily Life Dr. Benvikram Barman, Dr. Bhupendra Singh Banjare, Dr. Dolly Baghel, Dr. Dinesh Kumar Tandan, Mr. Naveen Verma, Mr. Rupendra Kumar Sahu, 2025-08-12 Chemistry in Daily Life is a comprehensive textbook designed to connect fundamental chemical principles with everyday applications. In this book, we have 23 chapters; each chapter explains with suitable figures and in the last section of each chapter, we have multiple types of questions with answers. This book explores how chemistry influences food, environment, health, and industry. It covers the composition and quality analysis of dairy products, beverages, food additives including artificial sweeteners, flavors, preservatives, colorants, and contaminants. Chapters on paints, dyes, pigments, and their industrial applications, air and water pollution, and observed finished topics like air pollutants, greenhouse gases, acid rain, and wastewater treatment. Readers gain understanding into water purification methods, the hydrologic cycle, and soil chemistry. The role of fertilizers and their classifications is explained alongside the chemistry of essential biomolecules like vitamins, carbohydrates, and oils. Medicinal chemistry is detailed with discussions on analgesics, antibiotics, antimalarials, and other drugs. The book also introduces concepts in chemical kinetics, thermodynamics, and energy systems including fuel cells, batteries, and solar energy. Finally, environmental challenges such as plastic waste, heavy metal pollution, are addressed with eco-friendly solutions. Rich in real-life examples, diagrams, and exercises, this book is perfect for students, educators, and anyone interested in how chemistry shapes our world.

Chemistry in Daily Life : Food, Medicine, Agriculture and Materials Dr. Kumar Sourav, 2025-06-20 Chemistry in Daily Life is a fascinating and insightful book title that immediately captures the essence of how deeply chemistry is embedded in our everyday experiences. Far from being confined to laboratories and textbooks, chemistry governs the basic phenomena of life from the air we breathe and the food we eat to the cleaning products we use and the emotions we feel. This book promises to bridge the gap between complex scientific principles and practical applications, making chemistry both accessible and relevant to readers of all backgrounds. In daily life, we often overlook the role of chemistry in simple tasks such as cooking, cleaning, or taking medicines. The Maillard reaction that browns our toast, the surfactants in our soaps, the preservatives in packaged food, or the pH-balanced shampoos are all marvels of chemical science. Even our body's metabolic functions are intricate biochemical reactions driven by enzymes, hormones, and neurotransmitters. A book titled Chemistry in Daily Life can enlighten readers about these everyday wonders and cultivate scientific curiosity. It can help students appreciate the subject better by showing them real-world applications, and it can serve as a handy reference for anyone who wants to understand how substances interact around and within them. By exploring such topics, this book not only enhances scientific literacy but also empowers individuals to make informed choices, whether it's selecting eco-friendly products, understanding nutrition labels, or comprehending the effects of drugs and cosmetics. Truly, this book holds the key to uncovering the invisible science that shapes our world.

The Tao of Chemistry and Life Eugene H. Cordes, 2009-06-04 Chemistry underlies life. This book establishes the relationship between the focal point

of chemistry the molecule and the key characteristics of living organisms The key is the interactions between small molecules and macromolecules leading to metabolic control memory and learning the senses and drug action **ISC Biology Book I for Class XI** Dr. P.S. Verma & Dr. B.P. Pandey, Well labelled illustrations diagrams tables figures and experiments have been given to support the text wherever necessary

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