

## RADIOACTIVITY WORKSHEET

### Questions:

1. Why did Marie Curie receive a second Nobel Prize? (1)  
**For her isolation of radium**
2. What happens to an atom with a nucleus that falls outside the band of stability? (3)  
**The nucleus will undergo radioactive decay to form a more stable nucleus**
3. What are the symbols used to identify a neutron, a proton and an electron in radioactive decay? (3)
4. Identify and explain THREE uses of radioisotopes. (6)
5. Explain what is meant by the statement 'Chlorine has two isotopes, chlorine-35 and chlorine-37'. (3)

### Fill in the Blanks:

6. Pierre and Marie Curie discovered the radioactive elements **polonium** and **radium**. (2)
7. Some isotopes are stable while other isotopes are **unstable** and will undergo **radioactive decay**. (2)
8. Three types of radiation include **alpha**, **beta** and **gamma**. (3)
9. During radioactive decay, the **nucleus** will emit particles of radiation in order to become more **stable**. (2)

# Chemistry Worksheet Questions 1 On Radioactivity Answers

**E Durkheim**



## **Chemistry Worksheet Questions 1 On Radioactivity Answers:**

*CBSE - ICSE Chemistry Part I* Chandan Sengupta, First Publication October 2021 Place of Publication Arabinda Nagar Bankura 722101 This workbook will provide an ample scope in getting exposed to the system of acquiring skills and competence related to the understanding of chemistry It also exposes the student to the concepts of chemistry for enabling the aspirant in acquisition of skills related to chemistry Some of the worksheets are prepared along with model answers Some other worksheets are meant for self assessment and evaluation purposes It is also observed that some of the topics are specific to the referred curriculum Some other topics are varyingly incorporated in other streams of study Culmination of more than two streams will enable the fellow student to cope up with the preparatory works meant for Olympiads and other competitive examinations *Chemistry* Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards **Learning Elementary Chemistry for Class 8 (A.Y.**

**2023-24) Onward** Dr. R. Goel, 2023-05-20 The series Learning Elementary Chemistry for Classes 6 to 8 has been revised strictly according to the latest curriculum The content of this series has been developed to fulfill the requirement of all the six domains Concepts Processes Applications Attitudes Creativity and World view of Science to make teaching and learning of Chemistry interesting understandable and enjoyable for young minds This series builds a solid foundation for young learners to prepare them for higher classes The main strength of the series lies in the subject matter and the experience that a learner will get in solving difficult and complex problems of Chemistry Emphasis has been laid upon mastering the fundamental principles of Chemistry rather than specific procedures Unique features of this series are The content of the book is written in a very simple and easy to understand language All the Key concepts in the curriculum have been systematically covered and graded in the text Each theme has been divided into units followed by thought provoking and engaging exercises to test the knowledge understanding and applications of the concepts learnt in that unit At the end of each theme a comprehensive theme assignment which is aligned with the guidelines provided in National Education Policy NEP 2020 is given Explanations illustrations diagrams experiments and solutions to numerical problems have been included to make the subject more interesting comprehensive and appealing Diagrams illustrations and text have been integrated to enhance comprehension Definitions and other important scientific information are highlighted Throughout the series

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