
Writing Balanced Equations CHEM 10 Review Worksheet

The equations supplied here are fairly simple and could be considered "Chem 10 level". They are provided to assist your review of some topics covered in Chp 3 of the Zumdahl textbook. An answer key is provided on the second page.

Writing Balanced Chemical Equations

Write balanced chemical equations for each of the following reactions:

1. Gaseous nitrogen dioxide reacts with water to produce nitric acid and gaseous nitrogen monoxide.
2. Rubidium metal (*s*) + water (*l*) → rubidium hydroxide (*aq*) + hydrogen (*g*)
3. Iodine (*s*) + chlorine (*g*) → iodine trichloride (*l*)
4. Sodium bicarbonate when heated will form solid sodium carbonate, carbon dioxide and water vapor.
5. Magnesium acetate (*aq*) + ammonium sulfide (*aq*) → magnesium sulfide (*s*) + ammonium acetate (*aq*).
6. C_6H_{14} (*l*) + oxygen (*g*) → carbon dioxide (*g*) + water (*l*) [C_6H_{14} is named hexane]
7. Iron(III) oxide (*s*) + carbon (*s*) → iron (*s*) + carbon dioxide (*g*)
8. Potassium carbonate (*aq*) + nitric acid (*aq*) → potassium nitrate (*aq*) + carbon dioxide (*g*) + water (*l*)
9. Ammonium nitrite (*s*) → nitrogen (*g*) + water (*l*)
10. Gaseous sulfur dioxide reacts with oxygen forming gaseous sulfur trioxide.
11. Cobalt(III) bromide (*aq*) reacts with pure chlorine forming pure bromine and cobalt(III) chloride (*aq*).
12. Silver (*s*) + S_8 (*s*) → silver sulfide (*s*) [S_8 is a naturally occurring form of sulfur]
13. Ammonia (*g*) + oxygen (*g*) → nitrogen monoxide (*g*) + water (*l*)
14. Aqueous barium hydroxide reacts with phosphoric acid forming water and solid barium phosphate.
15. Aluminum (*s*) + copper(II) sulfate (*aq*) → aluminum sulfate (*aq*) + copper (*s*)
16. When heated, solid lithium chlorate decomposes to produce solid lithium chloride and oxygen gas.
17. Benzene liquid (C_6H_6) burns in oxygen to form carbon dioxide and water (and heat).
18. Silicon dioxide (*s*) + hydrofluoric acid (*aq*) → silicon tetrafluoride (*g*) + water (*l*)

Chemistry Writing Balanced Equations Concept Review Answers

Clifford C. Houk, Richard Post



Chemistry Writing Balanced Equations Concept Review Answers:

Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

Let's Review Regents: Chemistry--Physical Setting Revised Edition Barron's Educational Series, Albert S. Tarendash, 2021-01-05 Barron's Let's Review Regents Chemistry gives students the step-by-step review and practice they need to prepare for the Regents Chemistry Physical Setting exam. This updated edition is an ideal companion to high school textbooks and covers all Chemistry topics prescribed by the New York State Board of Regents. Let's Review Regents Chemistry covers all high school level Chemistry topics and includes an extensive review of all topics on the test. Extra practice questions with answers. A detailed introduction to the Regents Chemistry course and exam. One actual recently released Regents Chemistry exam with an answer key.

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Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg, 2015-01-16 Ebook Chemistry: The Molecular Nature of Matter and Change. *Chemistry: Concepts and Problems* Clifford C. Houk, Richard Post, 1996-03-09 CHEMISTRY SECOND EDITION The fast, easy way to master the fundamentals of chemistry. Have you ever wondered about the differences between liquids, gases, and solids? Or what actually happens when something burns? What exactly is a solution? An acid? A base? This is chemistry: the composition and structure of substances composing all matter and how they can be transformed. Whether you are studying chemistry for the first time on your own, want to refresh your memory for a test, or need a little help for a course, this concise interactive guide gives you a fresh approach to this fascinating subject. This fully up-to-date edition of Chemistry Concepts and Problems has been tested

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Education Material Study, 1963 **Introductory Chemistry for Today** Spencer L. Seager, Michael R. Slabaugh, 1999-10-22
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