

## Unit 08 – Stoichiometry – Worksheet 1

Answer the following questions in the space provided at the left.

- 18 mol O<sub>2</sub> 1. How many moles of oxygen are produced from twelve moles of Potassium chlorate?



- 21 mol O<sub>2</sub> 2. How many moles of oxygen are produced from 14 moles of Potassium chlorate?



- 45 mol H<sub>2</sub>O 3. How many moles of water are produced from 22.5 moles of oxygen?



- 20 mol HCl 4. Calculate the number of moles of Hydrogen chloride produced from 10 moles of hydrogen.



- 21 mol Cl<sub>2</sub> 5. Calculate the number of moles of chlorine needed to form 14 moles of Iron (III) chloride.



# Chemistry Stoichiometry Answer Workbook Pages

**Leopold Gmelin**



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