

Chemistry: Ionic Bonding

Ionic Bonds are formed between metals and
nonmetals and involved the
formation of ions.

Charges that are close will repel each other, while
opposite charges will attract and bond
together.

In order to build an ionic compound, you must have both a
positive ion and a negative ion.

That positive ion will be formed from an atom that has a tendency to
lose electrons and will usually be a
metal.

A negative ion will be formed from an atom that has a tendency to
gain electrons and will usually be a
nonmetal.

Compounds metals and nonmetals
can bond together by forming an ionic bond.

When multiple ions pairs are attracted to each other it forms an ionic
crystal.

In ionic compounds the formula tells us the ratio between
elemental subscripts in the crystal.

Chemthink Ionic Charges Key

Karin Nielsen-Saines



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