#### APUSH Study and Reading Guides: Chapters 11-16 for The American Pageant (19th Edition)

#### Chapter 11: The Triumphs and Travails of Jeffersonian Democracy (1800-1812)

- Jefferson's Presidency: Thomas Jefferson's election in 1860 marked the first praceful transfer of power between political parties in the U.S. His presidency emphasized a limited federal government and reduced military presence while reversing some Pederalist policies.
- Louisiana Parchase: Jefferson's decision to purchase the Louisiana Territory from France in 1803 doubled the size of the U.S. and opened the way for westward expansion. This decision raised constitutional questions since the power to acquire new territory was not explicitly stated in the Constitution.
- Embargo Act (1887) in response to British and French attacks on American ships, Jefferson imposed the Embargo Act, which halted all American exports. This act was unpopular and economically damaging, especially to merchants and shipbudders.
- Harbury w. Madison: This landmark Supreme Court case established the principle of judicial review, allowing the Court to declare laws unconditutional, thus strengthening the judiciary.

Toolic	Details	Key Questions
jefferson's Presidency	First peaceful transfer of power; reduced government power.	How did pefferson's presidency reflect his political philosophy?
Louisiana Purchase	1803 purchase from France, doubled U.S. territory.	What were the effects and challenges of the Louisiana Funchase?
Embargo Act	Stopped all exports; economically damaging.	Why did jefferson impose the Embergo Act, and how did it backfire?
Marbury v. Madison	Established judicial review.	What is the significance of the Herbury v. Hedison cose?

#### Chapter 12: The Second War for Independence and the Opsurge of Nationalism (1912-1924)

 War of 1012. The U.S. entered into war with Britain over issues like impresement and trade restrictions. Though the war ended in a statemate.

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although the c cl bonds are rather polar the individual bond dipoles cancel one another in this symmetrical structure and does not have a net dipole moment cl 2 c ccl 2 exercise pageindex 3

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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the

molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry such as bf 3 trigonal planar ch 4 tetrahedral pf 5 trigonal bipyramidal and sf 6

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