

Kuta Software - Infinite Algebra 2

Solving Multi-Step Equations

Solve each equation

1) $4n - 2n = 4$

$$\frac{2n}{2} = \frac{4}{2}$$

$$n = 2$$

2) $-12 = 2 + 5v + 2v$

$$-12 = 2 + 7v$$

$$-2 = -2$$

$$\frac{-14}{7} = \frac{7v}{7}$$

$$-2 = v$$

3) $3 = x + 3 - 5x$

$$-4x + 3 = 3$$

$$-4x = 0$$

$$x = 0$$

4) $x + 3 - 3 = -6$

$$x - 3 = -6$$

$$x = -6$$

5) $-12 = 2 + 5v + 2v$

$$-12 = 2 + 7v$$

$$-14 = 7v$$

$$-2 = v$$

6) $-12 = 2 + 5v + 2v$

$$-12 = 2 + 7v$$

$$-14 = 7v$$

$$-2 = v$$

7) $-12 = 2 + 5v + 2v$

$$-12 = 2 + 7v$$

$$-14 = 7v$$

$$-2 = v$$

Kuta Software Infinite Algebra 2 End Behavior

RD Boyd



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 $2\text{NaClO}_3 \rightarrow 2\text{NaCl} + 3\text{O}_2$
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 $\text{Mg}(\text{OH})_2 + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2\text{O}$
 note all reactants and products require a coefficient of at least one
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 $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$

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 $2\text{C}_2\text{H}_2 + 5\text{O}_2 \rightarrow 4\text{CO}_2 + 2\text{H}_2\text{O}$
 if 1.0 mol of C_2H_2 is burned how many moles of CO_2 are formed
 $2.5 \text{ mol C}_2\text{H}_2 \rightarrow 5.0 \text{ mol CO}_2$
 if 1.0 mol of C_2H_2 is burned how many moles of oxygen gas are consumed
 2.5 mol O_2

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