



Kinetic Molecular Theory

I. MOLECULES IN MOTION:

A. Kinetic Molecular Theory (KMT) = the idea that particles of matter are always in motion and that this motion has consequences.

- 1) theory developed in the late 19th century to account for the behavior of the atoms and molecules that make up matter
- 2) based on the idea that particles in all forms of matter are always in motion and that this motion has consequences
- 3) can be used to explain the properties of solids, liquids, and gases in terms of the energy of particles and the forces that act between them

- 3 assumptions of KMT:

1. All matter (solid, liquid, gas) is made of particles (atoms, ions, molecules)

2. Particles are in constant motion (electrons move)

- Particles of a GAS travel in completely *random* motion
- Particles of a LIQUID appear to vibrate around *moving* points
- Particles of a SOLID appear to vibrate around *fixed* points



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