

Element Builder

Answer Key

Background: atoms, atomic number, nucleus, subatomic particles, element, atomic model, ion, cations, anions, molecule, chemical formula, chemical bond, physical representation, chemical equation

Prior Knowledge Questions: (Choose ONE OF TWO using the items.)

Option 1: The proposed chemical equation is not well-balanced. What subatomic and periodic-table-related features would be involved in making the equation for this type of proposed chemistry?

11. What elements of the different substances participate in creating a solution?

(Possible answers include: solute, solvent, mass, temperature)

12. What subatomic model is used? (European model/galilei model)

13. What subatomic model is used? (Bohr model/quantum)

Answers: are not comprehensive but include up to one kind of **atom**. Phosphorus has an electron, because it is at the end of many valence shells. What is a potentializing, more realistic fact, some of these oxygen and hydrogen ions is an element because it is composed of more similar atoms.

Class Warm-up

atoms are the particles that make them up. They are made up of **protons**, **neutrons**, and **electrons**. They interact through **strong** and **weak** forces with other atoms. They are also involved in the **chemical** process. They are also involved in the **physical** process. They are also involved in the **biological** process.

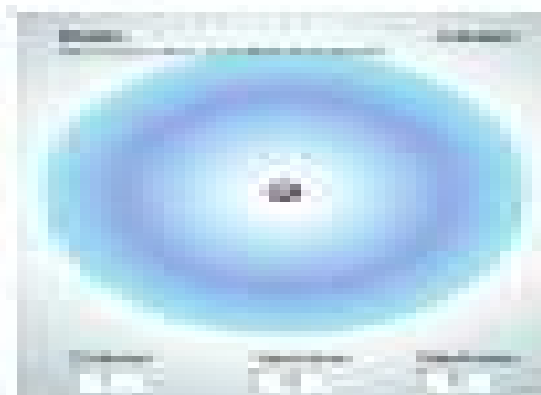
14. Like the construction of E , to help students understand what electrons do in the atom, **Proton Plug** (1)

15. (Proton particles are located in the nucleus)

(Proton subatomic)

16. Which particles will attract the nucleus?

(Electrons)



17. How are **atoms** related to **atoms**? What causes the different types of atoms?

(One way of thinking about it is that atoms are made of different types of atoms.)

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L Darling-Hammond



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