

Kinetics, Equilibrium and Thermodynamics: Practice Test

1. Multiple Choice

1. An enzyme is a biological catalyst. Which of the following is true?

- It provides an alternative mechanism.
- It provides an alternative mechanism.
- It provides a pathway of activation that reduces the activation energy of a reaction.
- It provides the (most) efficient of substrates that are not activated or transformed.
- It is not specific.

2. A study of chemical reactions with _____

- by determining the rate of reaction.
- by determining the activation energy.
- by determining the activation energy of the reaction.
- by determining the activation energy of the reaction.
- by determining the activation energy of the reaction.

3. Equilibrium is a state of _____

$$\begin{array}{l} \text{Step 1: } \text{A} + \text{B} \rightleftharpoons \text{C} + \text{D} \\ \text{Step 2: } \text{C} + \text{D} \rightleftharpoons \text{E} + \text{F} \end{array}$$

$$\text{Overall: } \text{A} + \text{B} \rightleftharpoons \text{E} + \text{F}$$

4. The equilibrium constant for the reaction is _____

$$K = \frac{[\text{C}][\text{D}]}{[\text{A}][\text{B}]} \quad K = \frac{[\text{E}][\text{F}]}{[\text{C}][\text{D}]} \quad K = \frac{[\text{E}][\text{F}]}{[\text{A}][\text{B}]} \quad K = \frac{[\text{C}][\text{D}]}{[\text{E}][\text{F}]} \quad K = \frac{[\text{A}][\text{B}]}{[\text{E}][\text{F}]}$$

5. A reaction is exothermic if _____

$$\begin{array}{l} \text{Step 1: } \text{A} + \text{B} \rightarrow \text{C} + \text{D} \\ \text{Step 2: } \text{C} + \text{D} \rightarrow \text{E} + \text{F} \end{array}$$

$$\text{Overall: } \text{A} + \text{B} \rightarrow \text{E} + \text{F}$$

6. The study of the effect of temperature on _____

$$K = \frac{[\text{C}][\text{D}]}{[\text{A}][\text{B}]} \quad K = \frac{[\text{E}][\text{F}]}{[\text{C}][\text{D}]} \quad K = \frac{[\text{E}][\text{F}]}{[\text{A}][\text{B}]} \quad K = \frac{[\text{C}][\text{D}]}{[\text{E}][\text{F}]} \quad K = \frac{[\text{A}][\text{B}]}{[\text{E}][\text{F}]}$$

7. The activation energy for a reaction is _____

- the energy barrier between a reactant and a product.
- the energy barrier between a reactant and a product.
- the energy barrier between a reactant and a product.
- the energy barrier between a reactant and a product.
- the energy barrier between a reactant and a product.



8. The study of chemical reactions is _____

$$K = \frac{[\text{C}][\text{D}]}{[\text{A}][\text{B}]} \quad K = \frac{[\text{E}][\text{F}]}{[\text{C}][\text{D}]} \quad K = \frac{[\text{E}][\text{F}]}{[\text{A}][\text{B}]} \quad K = \frac{[\text{C}][\text{D}]}{[\text{E}][\text{F}]} \quad K = \frac{[\text{A}][\text{B}]}{[\text{E}][\text{F}]}$$

9. The study of chemical reactions is _____

- The temperature is high enough to provide enough energy to overcome the activation energy.
- The temperature is high enough to provide enough energy to overcome the activation energy.
- The temperature is high enough to provide enough energy to overcome the activation energy.
- The temperature is high enough to provide enough energy to overcome the activation energy.

Kinetics And Equilibrium Practice Test

Neil D. Jespersen, Pamela Kerrigan



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