

Oxidation - Reduction Reactions

Redox Terminology—REVIEW CHAPTER 4!!!

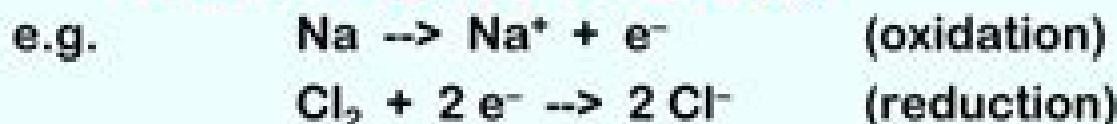
redox reaction -- electron transfer process



Overall process involves two Half Reactions:

oxidation -- loss of electron(s)

reduction -- gain of electron(s)



Related terms:

oxidizing agent = the substance that gets reduced (Cl_2)

reducing agent = the substance that gets oxidized (Na)

Oxidation and reduction always occur together so that there is no *net* loss or gain of electrons overall.

OILRIG

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Sebastian Brünink



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in Smart Materials for the Built Environment Cesare Sangiorgi, Filippo Ubertini, Anna Laura Pisello, 2018-12-10 This book is a printed edition of the Special Issue Recent Advances in Smart Materials for the Built Environment that was published in Materials **The Software Encyclopedia** ,1988 Science Workshop Series: Chemical changes Seymour Rosen, 2000

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preventive measurements against corrosion utilization of older materials in recent applications the use of new materials for existing equipment and more

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