

- Which best describes the current atomic theory?
 - atoms consist of electrons circling in a definite orbits around a positive nucleus.
 - atoms are composed of electrons in a cloud around a positive nucleus.
 - Atoms can easily be split, at which time they become radioactive.
 - an atom's mass is determined by the mass of its neutrons.
- What is the nuclear composition of uranium-235?
 - 92 electrons + 143 protons
 - 92 protons + 143 electrons
 - 143 protons + 92 neutrons
 - 92 protons + 143 neutrons
- Which best describes the relationship between subatomic particles in any neutral atom?
 - the number of protons equals the number of electrons
 - the number of protons equals the number of neutrons
 - the number of neutrons equals the numbers of electrons
 - the number of neutrons is greater than the number of protons
- What is the name of the compound PbO_2 ?
 - lead oxide
 - lead(II) oxide
 - lead oxide (I)
 - lead (IV) oxide
- What is the name of HCl(aq) ?
 - chloric acid
 - hydrochloric acid
 - hydrogen chloride
 - perchloric acid
- What is the formula for calcium nitrate?
 - CaNO_3
 - $\text{Ca}(\text{NO}_3)_2$
 - $\text{Ca}(\text{NO}_2)_2$
 - Ca_2N_2
- What is the correct formula for dinitrogen pentoxide?
 - N_4O
 - NO_2
 - N_2O_5
 - NO_4
- If the volume of 18.5 gram piece of metal is 2.35 cm^3 , what is the identity of the metal?
 - iron
 - lead
 - nickel
 - zinc

9. Which substance listed in the table is a liquid at 27°C ?

a. I b. II c. III d. IV

	Melting Point	Boiling Point
I	28°C	146°C
II	-16°C	28°C
III	20°C	146°C
IV	-90°C	14°C

- Which will increase the solubility of MOST solid solutes?
 - decreasing the temperature
 - decreasing the amount of solvent at constant temperature
 - increasing the amount of solute at constant temperature
 - increasing the temperature

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