

Acids, Bases, Solutions Study Guide

4 pages + 4-page Answer Key

chemistry

Physical Science

SOLUTIONS, ACIDS, AND BASES STUDY GUIDE

What is a heterogeneous mixture? Provide two examples.

Describe the characteristics of a saturated solution and an unsaturated solution.

Define the term solvent and provide an example.

What is solubility, and how is it measured?

Distinguish between concentrated and dilute.

What are the properties of acids?

What are the properties of bases?

What is the pH scale?

What is the pH of a neutral solution?

What are the properties of bases and how are they measured?

Explain the significance of the pH scale.

How does the concentration of an acid affect its pH?

What happens to the pH of a solution during neutralization?

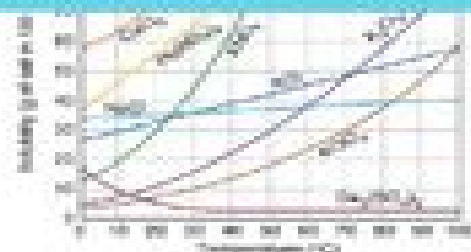
What happens to the pH of a solution when you add a strong acid?

What happens to the pH of a solution when you add a strong base?

What is the pH of a solution that is neutral?

What is the pH of a solution that is acidic?

What is the pH of a solution that is basic?



20. Describe how you would use the graph to determine the amount of KNO_3 that can be dissolved in 100 grams of water at 80°C.

22. Which salt has the highest solubility at 80°C?

ANSWER KEY INCLUDED



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M Woodhall



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