

D) in the lanthanoid and actinoid series

9. An atom has an atomic number of 18 and a mass number of 32. The number of electrons in the third orbit of the atom is

- A) 2 B) 3 C) 5 D) 6 E) 8

10. How many electrons, neutrons and protons would be found in an atom of carbon-14 (atomic # 6)?

- A) 6 electrons, 6 neutrons, 6 protons B) 6 electrons, 8 neutrons, 6 protons
C) 6 electrons, 6 neutrons, 8 protons D) 6 electrons, 6 neutrons, 6 protons

11. Which of the following compounds is NOT polar?

- A) ammonia B) citric acid C) methane D) none of these

12. Which of the following is a molecular compound?

- A) calcium oxide B) strontium chloride C) dinitrogen monoxide D) ammonium chloride

13. What type of bond is likely to form between the elements molybdenum, Mo and chlorine, Cl based on electronegativities?

- A) Non polar covalent B) Polar covalent C) ionic D) None of the above

14. Electrons are transferred from one element to another in a...

- A) polar covalent bond B) non-polar covalent bond C) ionic bond D) all of these

15. Metallic bonding is characterized by cations surrounded by a sea of:

- A) immobile valence electrons B) mobile protons
C) mobile neutrons D) mobile valence electrons

16. Which of the following has charged particles arranged in a highly ordered crystalline lattice?

- A) H_2 B) NaCl C) MgO D) CH_4 E) more than one of these

17. The correct name for $Mg(ClO_4)_2$ is _____.

- A) magnesium chlorate B) manganese chlorate C) magnesium chloroate
D) magnesium perchlorate E) manganese perchlorate

18. Al_2O_3 is named:

- A) dialuminum trioxide B) aluminum oxide C) aluminum (III) oxide D) aluminum trioxide

Gpb Chemistry Unit 11 Test Answers

J Dewey



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