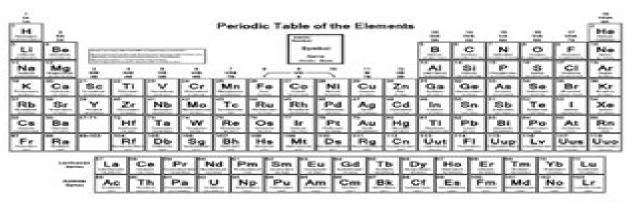
Chemistry Study Guide



Objective:

To understand the structure of an atom, and the nature + properties of subatomic particles 🛫

Matter:

Anything that has mass and occupies space 💥 🕥

Examples: air, gases, liquids, solidz!

Atoms:

Atoms are the smallest, indivisible particle &.

- Outer part → shell/orbit/energy level → carries the electronz (e-)
- Inner part → nucleus → houses the protons (p+) and neutronz (n)
- Outer & Inner part → together these are the subatomic particles

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p."	1	-1	Nucleus
Neutron	m	1	0	Nucleus
Electron	•	-0	-1	Orbit/Shell

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RJ Alexander

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In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is truly awe-inspiring. Enter the realm of "**Chemistry Study Guide Answers Solutions**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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