ssessment #3 Review	Name:
ou must show all work for	credit.
ie Date:	(No late packets will be accepted)
I. Balance Equation	ns.
Balance the equations	F II True
<ol> <li>Mg + ∠ H<sub>2</sub>O → M;</li> </ol>	$g(OH)_2 + H_2$
<ol> <li>Mg + ZNaBr → N</li> </ol>	MgBr <sub>2</sub> + ZNa
3. C <sub>3</sub> H <sub>8</sub> + 5 O <sub>2</sub> →3 C	
4. 4 P + 5 O <sub>2</sub> → 2 P <sub>2</sub> O <sub>5</sub>	1 1 1 1 CO 1 7 Ec/OH)
5. Fe <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub> + 6 KOH	I →3 K₂SO₄ + 2 Fe(OH)₃
<ol> <li>Fe<sub>2</sub>O<sub>3</sub> + 3 H<sub>2</sub> → F</li> </ol>	7
7 How many moles of	hydrogen gas react with 1 mole of Fe₂O₃ in reaction #6?
7. 11011 1112117	
rite the balanced equation	n for problems # 8 and 9:
<ol><li>Hydrogen gas and n</li></ol>	nitrogen monoxide combine to form water and nitrogen gas
C Man and make a	phate and calcium chloride are mixed they form calcium phosphate and
sodium chloride	2 Nas 704 13 Cacle - Cas (PON) = + 8 Na C
Sodium chloride	< Nastry 13 cacks - castronie actions
10 Using the following b	balanced equation, prove (or explain) the Law of Conservation of Mass.
$N_2 + 3H_2 \rightarrow 2NH_3$	
	Z N = lety Z = 1 ; [w]
	su no plants created or lost
The second second second second	equation shown in problem #10, explain what the coefficients represent.
11. Using the balanced	equation shown in problem #10, explain what the coemoleria represents
12. Using the balanced	equation shown in problem #10, list the products of the reaction.
	UKZ
13. Using the balanced	equation shown in problem #10, list the reactants of the reaction.
	1 12
II. Types of Reacti	ions
irections: Identify the type	e of reaction.
= synthesis	SD = single displacement C= combustion DD = double displacement
= decomposition	DD = double displacement
14. HCI + MgS	$O_4 \rightarrow MgCl_2 + H_2SO_4$
15. Cu₂(CO₃)₃ →	Cu + CO <sub>2</sub> + O <sub>2</sub>
16 C <sub>2</sub> H <sub>6</sub> + O <sub>2</sub>	→ CO <sub>2</sub> + H <sub>2</sub> O
	→ CO <sub>2</sub> + H <sub>2</sub> O
16. $C_2H_6 + O_2$	$\rightarrow$ CO <sub>2</sub> + H <sub>2</sub> O P <sub>2</sub> O <sub>5</sub>
16. $C_2H_6 + O_2$	$\rightarrow$ CO <sub>2</sub> + H <sub>2</sub> O P <sub>2</sub> O <sub>5</sub> CaCl <sub>2</sub> $\rightarrow$ NaCl + Ca(NO <sub>3</sub> ) <sub>2</sub>

## **Chemistry Test Review Unit 3 Answers**

**Tobias Bleicker** 

**Chemistry Test Review Unit 3 Answers:** 

The Enigmatic Realm of Chemistry Test Review Unit 3 Answers: Unleashing the Language is Inner Magic

In a fast-paced digital era where connections and knowledge intertwine, the enigmatic realm of language reveals its inherent magic. Its capacity to stir emotions, ignite contemplation, and catalyze profound transformations is nothing lacking extraordinary. Within the captivating pages of **Chemistry Test Review Unit 3 Answers** a literary masterpiece penned by way of a renowned author, readers attempt a transformative journey, unlocking the secrets and untapped potential embedded within each word. In this evaluation, we shall explore the book is core themes, assess its distinct writing style, and delve into its lasting effect on the hearts and minds of those who partake in its reading experience.

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