

CHEMISTRY

NEW!

Unit 13 Thermochemistry Learning Activities & Tests

13.1 Entropy of Reaction: Bond Energies

Calculate the enthalpy of reaction for a reaction using bond energies.

Worksheet: The enthalpy of reaction can be calculated using bond energies. Calculate the enthalpy of reaction for the following reaction.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$



For this reaction, calculate the enthalpy of reaction using bond energies.



Unit 13 Calculating: Entropy of Reaction (13.1)

Calculate the enthalpy of reaction for the reaction $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$ using bond energies. The enthalpy of reaction is the difference between the total bond energies of the products and the total bond energies of the reactants. The enthalpy of reaction is negative for exothermic reactions and positive for endothermic reactions.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$

Calculate the enthalpy of reaction for the reaction $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$ using bond energies.

- Calculate the total bond energies of the reactants.
- Calculate the total bond energies of the products.
- Calculate the enthalpy of reaction.

13.2 Heating and Cooling Curves

Interpret a heating and cooling curve for a substance. The heating and cooling curves for a substance show the temperature of the substance as it is heated or cooled. The curves show the phase changes of the substance and the energy required or released during the phase changes.

Worksheet: Interpret a heating and cooling curve for a substance. The heating and cooling curves for a substance show the temperature of the substance as it is heated or cooled.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$



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Worksheet: Interpret a heating and cooling curve for a substance. The heating and cooling curves for a substance show the temperature of the substance as it is heated or cooled.

Reaction: $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$



13.3 Energy Diagrams

Interpret an energy diagram for a reaction. An energy diagram for a reaction shows the energy of the reactants, the energy of the products, and the activation energy of the reaction.

- Interpret an energy diagram for a reaction.
- Calculate the activation energy of a reaction.
- Calculate the enthalpy of reaction.



Chemistry Unit 13 Thermochemistry Test Answers

**Christopher R. Hren, John T.
Moore, Peter J. Mikulecky**



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