Date:

Name:

 Complete, balance and classify the following equations as synthesis, decomposition, single replacement, double replacement, neutralization or combustion. (3 marks each)

a. Ca + Zn(NO<sub>1</sub>)<sub>2</sub> →

Reaction Type

b. C₂H6 + O₂ →

Reaction Type

c. Na<sub>2</sub>SO<sub>4</sub> + Ca(NO<sub>3</sub>)<sub>2</sub> →

Reaction Type

d. P<sub>4</sub>O<sub>10</sub> + H<sub>2</sub>O →

Reaction Type \_\_\_\_\_

e. Cl<sub>2</sub> + KI →

Reaction Type

f. Al + Cl<sub>2</sub> →

Reaction Type

g- Zn + Pb(ClO)₂ →

Reaction Type \_\_\_\_\_

Col<sub>3</sub> + (NH<sub>4</sub>)<sub>2</sub>CrO<sub>4</sub> →

Reaction Type

# **Classifying Reactions And Completing Equations**

**Heather Hattori, Richard H. Langley** 

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