

# Mole and Stoichiometry Mini-Lab

Calculations: Show **ALL** work (bridge/grid).

The GIVEN for calculations #1 and #2 should be measurements, not results from a previous calculations.

1. How many moles of aluminum are in the soda can? *UNKNOWN: mol Al*      *GIVEN: g Al (data table)*
2. How many atoms of aluminum are in the soda can? *UNKNOWN: atoms Al*      *GIVEN: g Al (data table)*
3. For an experiment, 2.5 moles of aluminum is needed. Since there is no measuring tool for moles, how would a researcher measure out this much aluminum? *UNKNOWN: g Al*      *GIVEN: 2.50 mol Al*

*Respond to question in complete sentence and show work for necessary calculation.*

# Chemistry Mini Lab 11 Apply Stoichiometry

**David Kirk**



## **Chemistry Mini Lab 11 Apply Stoichiometry:**

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