

Name: _____

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Key

Chemistry: Molar Mass and Percentage Composition

Calculate the molar masses and percentage composition of each of the following compounds. Show your work and always include units.

1. Ca_3P_2 $\overset{216.3 \text{ amu}}{3(40.08 \text{ amu})} = 66\% \text{ Ca}$ $100 - 66 = 34\% \text{ P}$
 $3(40.08) + 2(30.97 \text{ amu})$

2. $\text{Ca}(\text{OH})_2$ $\frac{40.08 \text{ amu Ca}}{40.08 \text{ amu} + 2(16.00) + 2(1.01)} = 54\% \text{ Ca}$ $\frac{2(16.00 \text{ amu O})}{74.1 \text{ amu}} = 43\% \text{ O}$ $\frac{2.02 \text{ amu H}}{74.1} = 3\% \text{ H}$

3. Na_2SO_4 $\frac{2(22.99 \text{ amu Na})}{2(22.99) + 32.07 + 4(16.00 \text{ amu O})} = 32\% \text{ Na}$ $\frac{32.07 \text{ amu S}}{142.05 \text{ amu}} = 23\% \text{ S}$ $\frac{4(16.00 \text{ amu O})}{142.05 \text{ amu}} = 45\% \text{ O}$

4. CaSO_4 $\frac{40.08 \text{ amu Ca}}{40.08 + 32.07 + 4(16.00 \text{ amu O})} = 29\% \text{ Ca}$ $\frac{32.07 \text{ amu S}}{136.15 \text{ amu}} = 24\% \text{ S}$ $\frac{4(16.00 \text{ amu O})}{136.15 \text{ amu}} = 47\% \text{ O}$

5. $(\text{NH}_4)_2\text{SO}_4$ $\frac{2 \times 14.01 \text{ amu N}}{2(14.01) + 8(1.01) + 32.07 + 4(16.00 \text{ amu O})} = 21\% \text{ N}$ $\frac{8(1.01 \text{ amu H})}{132.17 \text{ amu}} = 6\% \text{ H}$ $\frac{32.07 \text{ amu S}}{132.17 \text{ amu}} = 24\% \text{ S}$ $\frac{4(16.00 \text{ amu O})}{132.17 \text{ amu}} = 48\% \text{ O}$

6. $\text{Zn}_3(\text{PO}_4)_2$ $\frac{3(65.39 \text{ amu Zn})}{3(65.39) + 2(30.97) + 8(16.00 \text{ amu O})} = 51\% \text{ Zn}$ $\frac{2(30.97 \text{ amu P})}{386.11 \text{ amu}} = 16\% \text{ P}$ $\frac{8(16.00 \text{ amu O})}{386.11 \text{ amu}} = 33\% \text{ O}$

7. $\text{Mg}(\text{NO}_3)_2$ $\frac{24.31 \text{ amu Mg}}{(24.31) + 2(14.01) + 4(16.00 \text{ amu O})} = 16\% \text{ Mg}$ $\frac{2(14.01 \text{ amu N})}{148.33 \text{ amu}} = 19\% \text{ N}$ $\frac{4(16.00 \text{ amu O})}{148.33 \text{ amu}} = 65\% \text{ O}$

8. KCl $\overset{74.55 \text{ amu}}{39.10 \text{ amu K}} = 52\% \text{ K}$ $100 - 52 = 48\% \text{ Cl}$
 $39.10 + 35.45 \text{ amu}$

Answers:

- 182.3 g, 66.0% Ca, 34.0% P
- 74.1 g, 54.1% Ca, 43.2% O, 2.7% H
- 142.1 g, 32.4% Na, 22.6% S, 45.0% O
- 136.2 g, 29.4% Ca, 23.6% S, 47.0% O

- 132.1 g, 21.2% N, 6.1% H, 24.3% S, 48.4% O
- 386.2 g, 50.8% Zn, 16.1% P, 33.1% O
- 116.3 g, 20.9% Mg, 24.1% N, 55.0% O
- 74.6g, 52.4% K, 47.6% Cl

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R Bogdan



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Table of Contents Chemistry Mole And Equations Test Answer Key

1. Understanding the eBook Chemistry Mole And Equations Test Answer Key
 - The Rise of Digital Reading Chemistry Mole And Equations Test Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemistry Mole And Equations Test Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemistry Mole And Equations Test Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemistry Mole And Equations Test Answer Key
 - Personalized Recommendations
 - Chemistry Mole And Equations Test Answer Key User Reviews and Ratings

- Chemistry Mole And Equations Test Answer Key and Bestseller Lists
- 5. Accessing Chemistry Mole And Equations Test Answer Key Free and Paid eBooks
 - Chemistry Mole And Equations Test Answer Key Public Domain eBooks
 - Chemistry Mole And Equations Test Answer Key eBook Subscription Services
 - Chemistry Mole And Equations Test Answer Key Budget-Friendly Options
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 - Chemistry Mole And Equations Test Answer Key Compatibility with Devices
 - Chemistry Mole And Equations Test Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemistry Mole And Equations Test Answer Key
 - Highlighting and Note-Taking Chemistry Mole And Equations Test Answer Key
 - Interactive Elements Chemistry Mole And Equations Test Answer Key
- 8. Staying Engaged with Chemistry Mole And Equations Test Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemistry Mole And Equations Test Answer Key
- 9. Balancing eBooks and Physical Books Chemistry Mole And Equations Test Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemistry Mole And Equations Test Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chemistry Mole And Equations Test Answer Key
 - Setting Reading Goals Chemistry Mole And Equations Test Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chemistry Mole And Equations Test Answer Key
 - Fact-Checking eBook Content of Chemistry Mole And Equations Test Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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