

Part B-1

Answer all questions in this part.

Directions (31–50): For each statement or question, record on your separate answer sheet the number of the word or expression that, of those given, best completes the statement or answers the question. Some questions may require the use of the 2011 Edition Reference Tables for Physical Setting/Chemistry.

- 31 What is the number of electrons in an Al^{3+} ion? Low 5
Al +13p
- (1) 10 (3) 3
(2) 13 (4) 16

- 32 The valence electron of which atom in the ground state has the greatest amount of energy?

- ps. 6 (1) cesium ps. 5 (3) rubidium
ps. 2 (2) lithium ps. 3 (4) sodium

- 33 The numbers of protons and neutrons in each of four different atoms are shown in the table below.

Way #23
Protons and Neutrons in Four Different Atoms

Atom	Number of Protons	Number of Neutrons
A	8	8
D	9	9
E	9	10
G	10	10

Which two atoms represent isotopes of the same element?

- (1) A and D (3) E and D
(2) A and G (4) E and G

- 34 Which elements have the most similar chemical properties?

- Way #1
38
(1) boron and carbon
(2) oxygen and sulfur
(3) aluminum and bromine
(4) argon and silicon

- 35 Which element reacts with oxygen to form ionic bonds?

- Way #44
(1) calcium (3) chlorine
(2) hydrogen (4) nitrogen

- 36 The table below gives the atomic mass and the abundance of the two naturally occurring isotopes of chlorine.

Naturally Occurring Isotopes of Chlorine

Isotopes	Atomic Mass of the Isotope (u)	Natural Abundance (%)
^{35}Cl	34.97	75.76
^{37}Cl	36.97	24.24

Way #24
Which numerical setup can be used to calculate the atomic mass of the element chlorine?

- (1) $(34.97 \text{ u})(75.76) + (36.97 \text{ u})(24.24)$
(2) $(34.97 \text{ u})(0.2424) + (36.97 \text{ u})(0.7576)$
(3) $(34.97 \text{ u})(0.7576) + (36.97 \text{ u})(0.2424)$
(4) $(34.97 \text{ u})(24.24) + (36.97 \text{ u})(75.76)$

- Way #42
37 Which general trends in first ionization energy and electronegativity values are demonstrated by Group 15 elements as they are considered in order from top to bottom?

- (1) The first ionization energy decreases and the electronegativity decreases.
(2) The first ionization energy increases and the electronegativity increases.
(3) The first ionization energy decreases and the electronegativity increases.
(4) The first ionization energy increases and the electronegativity decreases.

- 38 An aluminum sample has a mass of 80.01 g and a density of 2.70 g/cm³. According to the data, to what number of significant figures should the calculated volume of the aluminum sample be expressed?

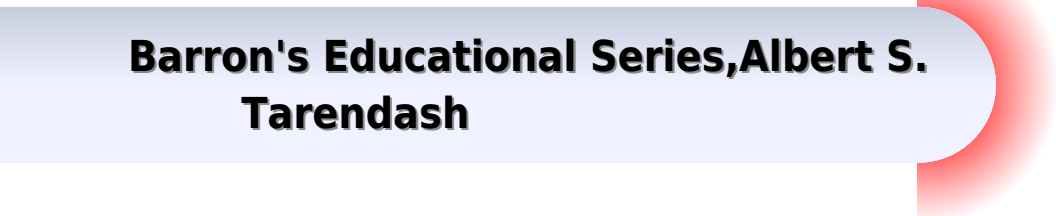
- (1) 1 (3) 3
(2) 2 (4) 4

$$D = \frac{M}{V} \quad ; \quad V = \frac{M}{D}$$

$$\frac{80.01 \text{ (4)}}{2.70 \text{ (3)}} \text{ (3)}$$

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**Barron's Educational Series, Albert S.
Tarendash**



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web jun 24 2019 employer summary paid time off is available to eligible jcpenny employees time off benefits are based on your employment status or average hours worked as determined during specific measurement periods

jcpenny employee kiosk net framework - Dec 28 2021

web the employee kiosk provides jcpenny workers with a safe method to access their paystubs manage their work schedules keep track of their staying vacation days examine the benefit programs they are signed up for and more

jcp associate kiosk portal net framework - Feb 27 2022

web only associates are permitted to use the jcp associate kiosk login portal associates can learn a great deal about their history and current status at the company through the portal things like pay stubs vacation days sick days personal days paid time off incentives assigned tasks weekly schedules and more can all be found in one place

jcpenny powerline former employees pdf prof - Mar 31 2022

web jcpenny associate kiosk is a site that allows employees to log in using their about the jcp associate kiosk is that the former employees of the associate jcp fa pdf employee handbook for jcpenny jcpennypowerline employee benefits login guide july 27th 2015 when sylva stoel a now former jcpenny employee showed up to work wearing

statement of business ethics penny ip llc - Jun 14 2023

web dear fellow associate on behalf of the senior management team we want to express our unwavering support for the ethical and business standards outlined in jcpenny s statement of business ethics the owners of jcpenny consisting of simon property group and brookfield asset management and our strategic partner authentic brands