

## Half-Yearly Examination—2013

## Class XI

(English Version)

Subject : Chemistry 1st Paper (Creative)

Time : 2 Hours 10 Minutes

Full Marks – 40

(Answer any four questions from the followings)

1. Read the following stem and answer the questions

|         |         |            |            |            |            |
|---------|---------|------------|------------|------------|------------|
| ${}_2A$ | ${}_4B$ | ${}_{12}C$ | ${}_{20}D$ | ${}_{24}E$ | ${}_{29}F$ |
|---------|---------|------------|------------|------------|------------|

Here A, B, C, D, E &amp; F are used symbolically not real symbol of elements.

- What is crystallization? 1
  - Why concentrated HCl is used in flame test? 2
  - Which of the above elements have equal number of electrons at outermost shell? 3
  - Two of the above elements have exceptional electronic configuration—explain logically with electronic configuration 4
2. An element has electronic configuration of  $(n-1)d^{10}ns^1$ . Here the value of n is 4 —
- What is steam distillation? 1
  - What do you mean by solubility of a salt is 70 at 20°C temperature? 2
  - Describe how will you identify the above ion? 3
  - Is the above element is transition element? If yes, then explain the formation of coloured compound of above element. 4
3. Read the following stem and answer the questions.

1 / IA      16 / VI A

|            |   |   |
|------------|---|---|
| 1st period | z |   |
| 2nd period |   | x |
| 3rd period |   | y |

- What is hybridization? 1
  - Which one is stronger between sigma and pi bond and why? 2
  - By which type of bond x and y form bonds with z — Explain. 3
  - The physical state of  $Z_2x$  and  $Z_2y$  are different — Explain logically. 4
4. A and B are two elements of atomic number 6 and 7 respectively. They reacts with hydrogen to form  $AH_4$  and  $BH_3$ .
- What is coordination bond? 1
  - Why Zn is d-block element but not transition element? 2
  - Explain the value of  $\angle HAH$  of  $AH_4$  by hybridization process. 3
  - Shape of  $AH_4$  and  $BH_3$  are different though they have similarly hybridized central atom.—explain logically. 4

5. 15 / VA      16 / VI A      17 / VII A

|            |   |   |   |
|------------|---|---|---|
| 2nd period | x | y | m |
| 3rd period |   |   | n |

# Chemistry Suggestion For Hsc Exam In 2015

**Maureen Slade, Roger Slade**



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