

Periodic Table Questions *[Place your answer in the blank to the left of the question number.]*

A. True or False [T or F](1-21)

- a) True
- b) False

- ___1. Properties of the elements are periodic functions of their atomic numbers.
- ___2. There are more nonmetallic elements than metallic elements.
- ___3. Metallic properties of the elements increase from left to right across a period.
- ___4. Metallic properties of the elements increase from top to bottom in a family of elements.
- ___5. Calcium is a member of the alkaline earth family.
- ___6. Iron belongs to the alkali metal family.
- ___7. Bromine belongs to the halogen family.
- ___8. Neon is a noble gas.
- ___9. Group A elements do not contain partially filled "d" or "f" orbitals.
- ___10. An atom of oxygen has a larger volume than an atom of lithium.
- ___11. An atom of sulfur is larger than an atom of oxygen.
- ___12. An atom of aluminum (Group 13) has five (5) electrons in its outer shell.
- ___13. Uranium is a transition element.
- ___14. The element with the ECN $1s^2 2s^2 2p^5$ belongs to Group 15.
- ___15. The element with the ECN $1s^2 2s^2 2p^6 3s^2 3p^1$ belongs to Group 13.
- ___16. A chemical family consists of one of the horizontal rows on the periodic table.
- ___17. Elements within a family will have similar electron structures and will show some similarities in their chemical properties.
- ___18. Group A elements are known as the representative elements.
- ___19. Going from left to right in any period, the representative elements show an increasing number of s and p electrons.
- ___20. The potassium ion is smaller than the potassium atom.
- ___21. The bromide ion is smaller than the bromine atom.

Chemistry Periodic Table Ii Answer Key

Arturo Cuomo



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