

Worksheet #3—Stoichiometry

Directions: Solve the following stoichiometry problems. Assume all gases are at STP. Don't forget to write out your balanced chemical equation. Make sure to show ALL work to receive FULL CREDIT!

1. Calculate the number of moles of oxygen required to react completely with 14.8 mol Al in the following equation: $4\text{Al}_{(s)} + 3\text{O}_{2(g)} \rightarrow 2\text{Al}_2\text{O}_{3(s)}$

14.8 moles Al	3 moles of O ₂	= <u>11.1 mol O₂</u>
	4 mole Al	

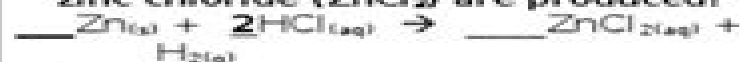
2. Calculate the moles of oxygen produced from the decomposition of 0.61 moles of potassium chlorate (KClO₃).



0.61 moles KClO ₃	3 moles of O ₂	=
	2 mole KClO ₃	

0.915 mol O₂

3. If 20.0g grams of zinc react with excess hydrochloric acid, how many grams of zinc chloride (ZnCl₂) are produced?



20 g Zn	1 mol Zn	1 mole ZnCl ₂	136 g ZnCl ₂
	65g Zn	1 mol Zn	1 mol ZnCl ₂

= 41.8 g ZnCl₂

4. How many grams of chlorine gas must be reacted with excess sodium iodide (NaI) if 10.0 grams of sodium chloride (NaCl) are needed



10 g NaCl	1 mol NaI	1 mol Cl ₂	71 g Cl ₂
	58.5g NaCl	2 mol NaCl	1 mol Cl ₂

= 6.07 g Cl₂

5. How many grams of copper are required to replace 4.00 grams of silver nitrate which are dissolved in water?



4.0g AgNO ₃	1 mol AgNO ₃	1 mole Cu	63.5 g Cu
	170g AgNO ₃	2 mol AgNO ₃	1 mol Cu

6. How many moles of hydrogen gas are needed to react with 15.1 g chlorine gas to produce hydrogen chloride gas?



15.1 g Cl ₂	1 mol Cl ₂	1 mole H ₂
	71 g Cl ₂	1 mole Cl ₂

= 0.21 moles H₂

Chemistry Stoichiometry Problems 1 Answers

**Robert J. Salm, Robert J. Salm Robert J.
Salm Ph.D.**

Chemistry Stoichiometry Problems 1 Answers:

Dr. Salm's Chemistry Problem Drill Book Robert J. Salm, Robert J. Salm Robert J. Salm Ph.D., 2005 This drill book contains many common problem types that are asked in General Chemistry classes in High School and College This work will give you practice with the major problem types as you prepare for finals and standardized tests

Microbiology (Questions and Answers), 5e Purshotam Kaushik & Kirti Kaushik, 2022 Microbiology is an engaging textbook presenting balanced and comprehensive account of major areas of microbiology in the form of questions and answers This question answer approach to present complex topics and theories of microbiology regarding cellular and non cellular microorganisms microbial genetics and molecular biology in higher plants and animals makes the subject interesting and easily comprehensible for the students

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Barron's Science 360: A Complete Study Guide to Chemistry with Online Practice Barron's Educational Series, Mark Kernion, Joseph A. Mascetta, 2021-09-07 provides a complete guide to the fundamentals of chemistry Page 4 of cover

CLEP® Chemistry Book + Online Kevin Reel, 2013-01-16 REAs new CLEP Chemistry with Online Practice Tests comes with 2 full length practice tests in the book and the same 2 tests with a length diagnostic test in timed format with instant scoring Online The comprehensive review covers all official topics Structure of Matter States of Matter Reaction Types Equations and Stoichiometry Equilibrium Kinetics Thermodynamics Descriptive Chemistry Experimental Chemistry Also includes test taking tips and study strategies

Chemistry, Student Study Guide James E. Brady, Fred Senese, 2008-01-28 The image on the front cover depicts a carbon nanotube emerging from a glowing plasma of hydrogen and carbon as it forms around particles of a metal catalyst Carbon nanotubes are a recently discovered allotrope of carbon Three other allotropes of carbon buckyballs graphite and diamond are illustrated at the left as is the molecule methane CH_4 from which nanotubes and buckyballs can be made The element carbon forms an amazing number of compounds with structures that follow from simple methane found in natural gas to the complex macromolecules that serve as the basis of life

on our planet The study of chemistry also follows from the simple to the more complex and the strength of this text is that it enables students with varied backgrounds to proceed together to significant levels of achievement

Chemistry Neil D. Jespersen, Alison Hyslop, 2021-11-02 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

Intelligent Tutoring Systems Mitsuru Ikeda, Kevin Ashlay, Tak-Wai Chan, 2006-06-21 This book constitutes the refereed proceedings of the 8th International Conference on Intelligent Tutoring Systems ITS 2006 held in Jhongli Taiwan June 2006 The book presents 67 revised full papers and 40 poster papers together with abstracts of 6 keynote talks organized in topical sections on assessment authoring tools bayesian reasoning and decision theoretic approaches case based and analogical reasoning cognitive models collaborative learning e learning and web based intelligent tutoring systems and more

The Chemist's Manual Henry Augustus Mott, 1877

Mapping College Chemistry Stephen DeMeo, 2019-03-01 This text is a chemistry problem solving resource appropriate for teachers and their students who are enrolled in high school Advanced Placement Chemistry or in a first year college General Chemistry course The book incorporates a chemistry problem solving plan one that uses an innovative graphic organizer strategy The strategy successfully evaluated with students combines problem solving processes with chemical concepts that will allow students to solve the most common and difficult problems encountered in the first year of chemistry Topical problem solving will focus on limiting reactant stoichiometry identifying types of chemical reactions equilibrium acid base equilibria and electrochemistry Why would this resource be of interest to chemistry students To be successful to get into a well known college medical school physical therapy or graduate program often requires that students get an A in your pre requisite Introductory General Chemistry course To make matters worse many college professors feel that only a few students should get A grades and therefore they give difficult exams that many students fail this is the weeding out process that every pre health student is apprehensive about To succeed in this competitive environment entails not just studying harder or longer it means re organizing textbook content so that it is meaningful to the student This is the first text of its kind to employ a

reliable research based strategy that incorporates a decision based visual tool to solve chemistry textbook problems ones that can make or break a career **CliffsNotes Chemistry Practice Pack** Charles Henrickson,2010-02-08 About the Contents

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Carson-Dellosa Publishing,2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards **Modern Methods in Kinetics** C.H. Bamford,R.G. Compton,C.F.H. Tipper†,1983-01-01 Modern Methods in Kinetics Chemistry, Problem-Solving Worktext James E. Brady,Frederick Senese,2002-06-13 The images on the cover call

attention to the relationship between macro observations and the intimate structure of chemical substances and the changes both chemical and physical that they undergo Fireworks One of the ingredients is phosphorus a molecular form of which is believed to consist of linked tetrahedra of phosphorus atoms The chemical reaction of phosphorus with oxygen is partly responsible for the spectacular show of light Carbon The element is found in several forms including the familiar diamond and another recently discovered sooty substance that consists of soccer ball shaped molecules often referred to as buckeyballs Diamond is not the most stable form of carbon and is created from other forms of carbon at high temperatures and pressures deep within the earth Acetylene torch Cutting steel is possible because of the intense heat generated by the chemical reaction of acetylene with oxygen a reaction between molecules of C_2H_2 and O_2 to give CO_2 and H_2O Hot air balloon The air that helps it rise is heated by the combustion of molecules of propane each composed of three carbon and eight hydrogen atoms Stormy weather The evaporation of water serves to store energy provided by the sun Subsequent condensation of the water vapor releases this energy and is the basis of all the weather systems on our planet Resources

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