Chemistry Review						Name			
1.	What is the correct electron dot diagram for (1) Al: (2) Al:			agram for	e the aluminum atom in the gro				
2.	Which pair of ator (1) ∞X ²²⁸ and ∞X ²	ns are is	otopes o (2) 912	d element c ²²⁷ and ₉	X? X ²²⁰	(3) ₉₀ X ²³⁸	and 95X ²²⁷	@ 50X ²²⁷ and 50X ²²⁶	
) 8(A binary compound of sodium is (1) sodium chlorate (2) sodium chlo				orite	(3) sodium	n perchlorate	e (4) sodium chloride	
4.	When the vapor pressure in an open container equal boil (2) freeze (3) crystallize					als the atmospheric pressure, the liquid will (4) melt			
350	When the equation smallest whole nu (1) 1 (2) 2						H ₂ (g) is c	orrectly balanced using	
6.	An ionic bond forms between atoms of (1) I and Cl				and CI	(4) H and CI			
7.	The greatest absorption of energy occurs as an electron moves from (3) 1s to 3s (2) 3p to 3s (3) 4d to 4s (4) 4s to 3p							to 3p	
8.	What is the total n	umber o (2) 2	f valano	e electron	s in an	atom with a	total of 13	electrons?	
9.	Which of the following has the least mass? (1) proton (2) neutron (3) hy				drogen nucl	cus	electron		
10.	What is the total n	umber o	f sublev (3) 3	els in the	(4) 4	l principal es	ergy level?		
11.	Which of the follo (1) Al	wing ato (2) Rb	onos has	the greate (3) I	est bond	ency to gain	electrons?		
32	Which is the formu (1) NH ₄ CH ₅ O					(3) NaCH	scoo	(4) CH ₃ COONH ₄	
12.	What type of bond is present in copper wire? (1) covalent (2) ionic (3) el-					ectrovalent		en metallic	
13.	Which type of bond is present in a water molecule? Dolar covalent (2) nonpolar covalent						(4) m	etallic	
14.	As sodium reacts with fluorine to form the compound NaF, each sodium atom will (1) gain 1 electron (2) gain 2 electrons (3) lose 1 electron (4) lose 2 electrons								
150	(I) hydrosulfaric a	name o	f the cor (2) by	npound v	vith the	formula, H ₂ id (3) sulfuri	SO ₄ ? ic sold	(4) sulfurous acid	

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