

Chemistry – Unit 7 Worksheet B

Write balanced chemical equations for the following reactions.

1. Ammonia (NH_3) reacts with hydrogen chloride to form ammonium chloride.



2. Calcium carbonate decomposes upon heating to form calcium oxide and carbon dioxide.



3. Barium chloride reacts with water to form barium hydroxide.



4. Acetylene (C_2H_2) decomposes to form ethylene (C_2H_4) and carbon monoxide.



5. Zinc reacts with copper (II) chloride to form zinc chloride and copper.



6. Calcium sulfate decomposes upon heating to form calcium oxide and sulfur dioxide.



7. Iron reacts with sulfuric acid (H_2SO_4) to form iron (II) sulfate and hydrogen gas.



8. A mixture containing carbon monoxide, CO , decomposes to form carbon, C , and nitrogen gas.

9. Phosphorus, P_4 , is burned in oxygen to form phosphorus pentoxide with water gas.

Chemistry Unit 7 3 V2

**Educart, Bharat Panchal, Abhishek
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Chemistry Unit 7 3 V2:

Solutions, Phase equilibrium, Conductance, Electrochemistry and Functional Group Organic Chemistry II - Laboratory

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