

Name Solutions

Date _____ Period _____

Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



| Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|-------|-------------------|-------------------------------------|----------------------------|
| Li | 1 ve | transfers $1e^-$ to F | Li^+ |
| F | 7 ve | accepts $1e^-$ from Li | F^- |

| Reactions | Atoms | Valence electrons | Electron transfer from/to each atom | Ions formed in the product |
|-------------------------------|-------|-------------------|-------------------------------------|----------------------------|
| 1) $Li + Cl \Rightarrow LiCl$ | Li | 1 ve | donates $1e^-$ | Li^+ |
| $Li + Cl \Rightarrow LiCl$ | Cl | 1 ve | accepts $1e^-$ | Cl^- |
| 2) $Ca + O \Rightarrow CaO$ | Ca | 2 ve | donates $2e^-$ | Ca^{+2} |
| $Ca + O \Rightarrow CaO$ | O | 6 ve | accepts $2e^-$ | O^{+2} |
| 3) $Be + F \Rightarrow BeF_2$ | Be | 2 ve | donates $2e^-$ to each F | Be^{+2} |
| $Be + F \Rightarrow BeF_2$ | F | 7 ve | each F accepts $1e^-$ each | F^- |
| 4) $Mg + S \Rightarrow MgS$ | Mg | 2 ve | donates $2e^-$ to S | Mg^{+2} |
| $Mg + S \Rightarrow MgS$ | S | 6 ve | accepts $2e^-$ | S^{+2} |
| 5) $K + F \Rightarrow KF$ | K | 1 ve | donates $1e^-$ | K^+ |
| $K + F \Rightarrow KF$ | F | 7 ve | accepts $1e^-$ | F^- |

Chemquest 20 Advanced Ionic Bonding Worksheet

Primrose Kitten

A decorative red circular graphic with a gradient, appearing as a partial circle or a stylized 'C' shape, located to the right of the 'Primrose Kitten' text.

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1. Understanding the eBook Chemquest 20 Advanced Ionic Bonding Worksheet
 - The Rise of Digital Reading Chemquest 20 Advanced Ionic Bonding Worksheet
 - Advantages of eBooks Over Traditional Books
2. Identifying Chemquest 20 Advanced Ionic Bonding Worksheet
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chemquest 20 Advanced Ionic Bonding Worksheet
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chemquest 20 Advanced Ionic Bonding Worksheet
 - Personalized Recommendations
 - Chemquest 20 Advanced Ionic Bonding Worksheet User Reviews and Ratings
 - Chemquest 20 Advanced Ionic Bonding Worksheet and Bestseller Lists
5. Accessing Chemquest 20 Advanced Ionic Bonding Worksheet Free and Paid eBooks
 - Chemquest 20 Advanced Ionic Bonding Worksheet Public Domain eBooks
 - Chemquest 20 Advanced Ionic Bonding Worksheet eBook Subscription Services
 - Chemquest 20 Advanced Ionic Bonding Worksheet Budget-Friendly Options

6. Navigating Chemquest 20 Advanced Ionic Bonding Worksheet eBook Formats
 - ePub, PDF, MOBI, and More
 - Chemquest 20 Advanced Ionic Bonding Worksheet Compatibility with Devices
 - Chemquest 20 Advanced Ionic Bonding Worksheet Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chemquest 20 Advanced Ionic Bonding Worksheet
 - Highlighting and Note-Taking Chemquest 20 Advanced Ionic Bonding Worksheet
 - Interactive Elements Chemquest 20 Advanced Ionic Bonding Worksheet
8. Staying Engaged with Chemquest 20 Advanced Ionic Bonding Worksheet
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chemquest 20 Advanced Ionic Bonding Worksheet
9. Balancing eBooks and Physical Books Chemquest 20 Advanced Ionic Bonding Worksheet
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chemquest 20 Advanced Ionic Bonding Worksheet
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Chemquest 20 Advanced Ionic Bonding Worksheet
 - Setting Reading Goals Chemquest 20 Advanced Ionic Bonding Worksheet
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Chemquest 20 Advanced Ionic Bonding Worksheet
 - Fact-Checking eBook Content of Chemquest 20 Advanced Ionic Bonding Worksheet
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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