

9. $\text{N}_2\text{O}_4(\text{g}) \rightleftharpoons 2 \text{NO}_2(\text{g})$ $.10$ $.02$ @ 25°C
 At 25°C , 0.11 mole of N_2O_4 reacts to form 0.10 mol of N_2O_4 and 0.02 mole of NO_2 . At 90°C , 0.11 mole of N_2O_4 forms 0.050 mole of N_2O_4 and 0.12 mole of NO_2 . From these data we can conclude $.05$ $.12$ @ 90°C
- N_2O_4 molecules react by a second order rate law.
 - N_2O_4 molecules react by a first order rate law.
 - the reaction is ~~exothermic~~ *endo*.
 - N_2O_4 molecules react faster at 25°C than at 90°C .
 - the equilibrium constant for the reaction above increases with an increase in temperature. *more products at 90°C than 25°C*

10. For the equilibrium system
 $\text{H}_2\text{O}(\text{g}) + \text{CO}(\text{g}) \rightleftharpoons \text{H}_2(\text{g}) + \text{CO}_2(\text{g})$ *+ heat*
 $\Delta H = -42 \text{ kJ/mol}$
 K_c equals 0.62 at 1260 K . If 0.10 mole each of H_2O , CO , H_2 and CO_2 (each at 1260 K) were placed in a 1.0-Liter flask at 1260 K , when the system came to equilibrium...

	The temperature would	The mass of CO would
a)	decrease	increase
b)	decrease	decrease
c)	remain constant	increase
d)	increase	decrease
e)	increase	increase

$$Q = \frac{(.10)(.10)}{(.10)(.10)} = 1 \quad K_c = .62$$

$Q > K_c$ *will shift to*
To use up products and make reactants
Temp $\downarrow$ $\text{CO} \uparrow$

11. For the reaction system,
 $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightleftharpoons 2\text{NH}_3(\text{g}) + \text{heat}$
 the conditions that would favor maximum conversion of the reactants to products would be *fewer mols of gas on product side*
- high-temperature and high pressure
 - high-temperature, pressure unimportant
 - high-temperature and low pressure
 - low temperature and high pressure
 - low temperature and low pressure
12. Solid HgO , liquid Hg , and gaseous O_2 are placed in a glass bulb and are allowed to reach equilibrium at a given temperature.
 $2\text{HgO}(\text{s}) \rightleftharpoons 2\text{Hg}(\text{l}) + \text{O}_2(\text{g})$ *heat* $\Delta H = +43.4 \text{ kcal}$
 The mass of HgO in the bulb could be increased by *shift left*
- adding more Hg . *NO CHANGE*
 - removing some O_2 . *shift right*
 - reducing the volume of the bulb *fewer mols of gas on reactant side*
 - increasing the temperature. *shift right*
 - removing some Hg . *NO CHANGE*

Answers: (Please use *CAPITAL* letters)

1.	E
2.	D
3.	C
4.	B
5.	D
6.	D

7.	D
8.	A
9.	E
10.	A
11.	D
12.	C

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Brendan G. Carr



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web dear church members greetings in jesus name i m writing this letter with much love in my heart to extend my appreciation to each and every one of you you don t how much you have touched my heart if i had a chance i will send each one of you a letter of appreciation on behalf of my family accept our appreciation for your prayer and support

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web oct 13 2020 sub letter to welcome retreat participants dear name of the recipient we are very happy and pleased to welcome you to our mention the name of the church church as the new retreat participants it is our great pleasure to have you with us as the retreat participants we are happy to get the opportunity to pray before our lord with

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web letter of thanks from pastor to church janitor dear name i hope this letter finds you in the best of health and spirits i am writing to express my heartfelt gratitude for your unwavering faithfulness and the exceptional job you have been doing as our church janitor your commitment to maintaining the cleanliness and orderliness of our

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web thanks for a visit i really appreciate your visiting me in the hospital when i am feeling ill it s nice to have special people like you visit me your prayer gave me peace about my health worries thanks for coming to my house and visiting me talking and praying with me meant a lot and brightened my day

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optional reiterate your thanks or offer a compliment or other friendly comment

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web oct 13 2020 subject letter of encouragement for catholic retreat dear name of the person this letter is to encourage you to a catholic retreat i want to start by saying how proud i am of you for taking the initiative to be in this divine of being catholic retreat putting god first takes a lot of courage and strength