CHAP # 01 Sticenionetry :-> Check word - measurement of elements. "study of Teactants and product quantitively." 2 H26 + O2 -> 2 H20 2 molecules of "H" + 1 molecul + 10, -> 2 molecules of +1+0 2 mol of H, + 1 mol . 10, -> 2 mol of H20 4 g H, + 32 g O2 -> 2 x 18 g H20 36 g redunt -> 36 g product 2H2 + 02 -> 2H200 2 (22.4 dm3) mt 22.4 dm3 Or > 2 (22.4 dm1 Hzo) MCGS:7 2H2 + O2 - 2H20

2pmolH2 + Imol O2 - 2 mol H20 4 mal of Hz + 2mol Or -> 4 mol Hz6 1 mol of H2 + 6.5 mol Oz -> 1 mol H20 4 g H2 + 32 g O2 -> 36g H20 2 g H2 + 16 g Oz -> 18 g H20 -10

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