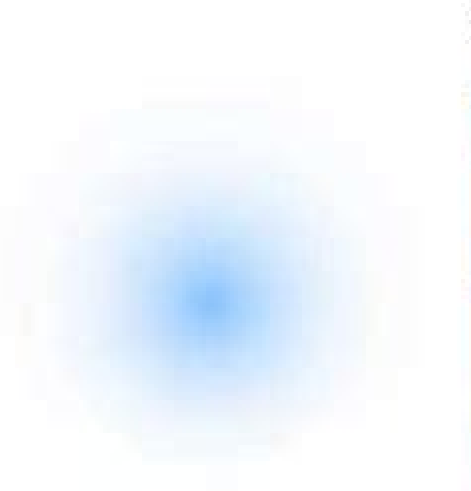
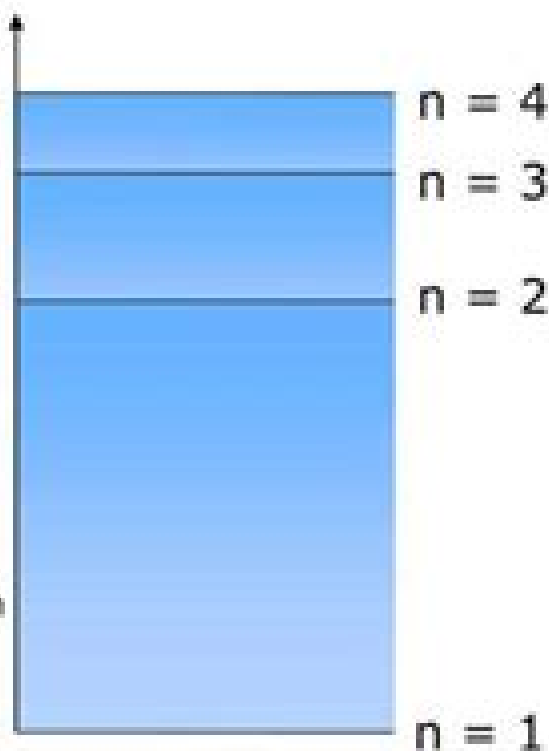


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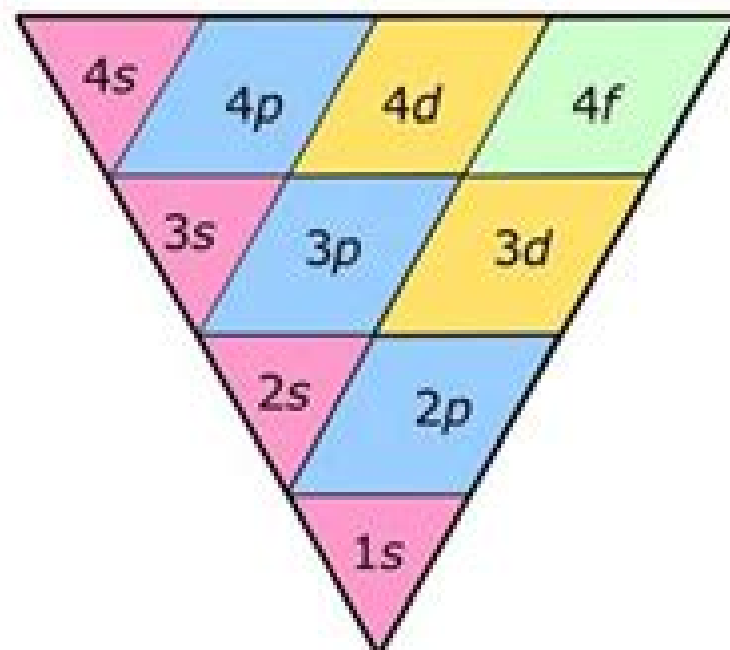


An orbital for a hydrogen atom. The intensity of the dots shows that the electron spends more time closer to the nucleus.



The first four principal energy levels in the hydrogen atom. Each level is assigned a principal quantum number n .

Sublevel designation



The types of orbitals on each of the first four principal energy levels.

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