

Name: _____ Date: _____ Block: _____

Ionic bonding practice

How many valence electrons do each of the following elements have?

- 1) carbon _____
- 2) selenium _____
- 3) xenon _____
- 4) potassium _____

Which of the following ions are likely to be formed? Circle one:

- | | | | | | |
|-------------|-----|----|---------------|-----|----|
| 5) N^{+5} | yes | no | 8) Al^{+2} | yes | no |
| 6) He^{+} | yes | no | 9) P^{-3} | yes | no |
| 7) F^{-1} | yes | no | 10) Mg^{+2} | yes | no |

Write the formula which result from combining the following ions:

- 11) K^{+1} and Br^{-} _____
- 12) Li^{+1} and O^{-2} _____
- 13) Ca^{+2} and Cl^{-1} _____
- 14) Al^{+3} and S^{-2} _____
- 15) Fe^{+2} and S^{-2} _____
- 16) Na^{+1} and NO_3^{-1} _____
- 17) Na^{+1} and SO_4^{-2} _____
- 18) NH_4^{+1} and CO_3^{-2} _____
- 19) Fe^{+3} and CrO_4^{-2} _____
- 20) Al^{+3} and PO_4^{-3} _____

Covalent And Ionic Bonding Questions And Answers

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In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Covalent And Ionic Bonding Questions And Answers**," written by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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