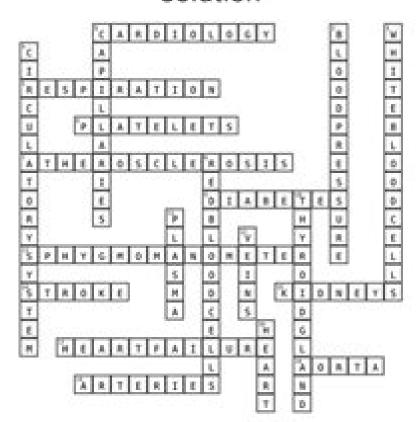


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calculate the number of moles from each of the following a 6 95 x 10 23 atoms of cu b 7 12 x 10 21 molecules of co 2 c 3 33 x 10 22 ions of pb 4 d 7 00 x 10 23 formula units of lino 3

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the quantity of heat for a process is represented by the letter q the sign of q for an endothermic process is positive because the system is gaining heat a chemical reaction or physical

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one milligram is equal to 0 001 grams 790 drag each number to the correct location sort the numbers based on the number of significant figures they have one significant figure 1 000

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the quantities commonly encountered in chemical problems include the number of moles of a substance the number of atoms molecules or formula units of a substance and the mass in

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it is always a good idea to estimate the answer before you do the actual calculation in this case the mass given 35 00 g is less than the molar mass so the answer should be less than 1

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