

3.4 Cycles of Matter

Lesson Objectives

-  Describe how matter cycles among the living and nonliving parts of an ecosystem.
-  Describe how water cycles through the biosphere.
-  Explain why nutrients are important in living systems.
-  Describe how the availability of nutrients affects the productivity of ecosystems.

Lesson Summary

Recycling in the Biosphere _____, unlike energy, is _____ within and between ecosystems. Elements pass from one organism to another and from one part of the biosphere to another through _____, which are closed loops powered by the flow of energy.

The Water Cycle Water moves between the _____, the _____, and _____.

- ▶ Evaporation is the process in which water changes from a liquid to a gas.
- ▶ Transpiration is the process in which water evaporates from the leaves of plants. Other steps in the water cycle include _____.

Nutrient Cycles The chemical substances that an organism needs to survive are called _____. Like water, nutrients pass through organisms and the environment.

- ▶ **Carbon Cycle:** Carbon is a key ingredient of all _____. Processes involved in the carbon cycle include photosynthesis and human activities such as burning.
- ▶ **Nitrogen Cycle:** Nitrogen is needed by all organisms to _____. Processes involved in the nitrogen cycle include _____ and _____.
 - In **nitrogen fixation**, certain _____ convert _____ into _____.
 - In **denitrification**, other soil _____ nitrogen compounds called _____ back into _____.
- ▶ **Phosphorus Cycle:** Phosphorus is needed for molecules such as _____ and _____. Most of the phosphorus in the biosphere is _____ and _____. Stored phosphorus is gradually released into water and soil, where it is used by organisms.

Nutrient Limitation A nutrient that, in short supply, can limit the productivity of an ecosystem is called a _____.

Cycles Of Matter Activity Answer Key Biology

L Darling-Hammond



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web procedure 1 find the mass of a clean dry 100 ml beaker and record 2 measure out 8 00 grams of copper ii sulfate onto a

piece of filter paper taking the filter paper's mass into account add these to the beaker 3 measure 50.0 ml of water in the cylinder and add it to the beaker 4

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web under the finalize of the experiment it was found that iron was the limiting reactant and iron ii sulfate was a product of the reaction between 7 0002g away cooper ii sulfate and 2 0101g iron metal which produced 2 3037g of copper product with a percent yield of 100 72 introduction

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