

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers?

Flora can give Frank her blue marker, and in return, Frank can give Flora his red marker. This way, both will have a full set of eight markers. _____

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

Otto can give Olivia his black and brown markers, and Olivia can give Otto her purple and green markers. This exchange would help both of them complete their sets with all six markers. _____

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?
2. Click **Pause** (). Drag a valence electron from the left atom to the right atom. Click **Play**.

What happens? _____

Covalent Bond Gizmo Worksheet Answers

IM Harris



Covalent Bond Gizmo Worksheet Answers:

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Table of Contents Covalent Bond Gizmo Worksheet Answers

1. Understanding the eBook Covalent Bond Gizmo Worksheet Answers
 - The Rise of Digital Reading Covalent Bond Gizmo Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Bond Gizmo Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Bond Gizmo Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Bond Gizmo Worksheet Answers
 - Personalized Recommendations
 - Covalent Bond Gizmo Worksheet Answers User Reviews and Ratings
 - Covalent Bond Gizmo Worksheet Answers and Bestseller Lists
5. Accessing Covalent Bond Gizmo Worksheet Answers Free and Paid eBooks
 - Covalent Bond Gizmo Worksheet Answers Public Domain eBooks
 - Covalent Bond Gizmo Worksheet Answers eBook Subscription Services
 - Covalent Bond Gizmo Worksheet Answers Budget-Friendly Options
6. Navigating Covalent Bond Gizmo Worksheet Answers eBook Formats

- ePub, PDF, MOBI, and More
 - Covalent Bond Gizmo Worksheet Answers Compatibility with Devices
 - Covalent Bond Gizmo Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Covalent Bond Gizmo Worksheet Answers
 - Highlighting and Note-Taking Covalent Bond Gizmo Worksheet Answers
 - Interactive Elements Covalent Bond Gizmo Worksheet Answers
8. Staying Engaged with Covalent Bond Gizmo Worksheet Answers
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covalent Bond Gizmo Worksheet Answers
9. Balancing eBooks and Physical Books Covalent Bond Gizmo Worksheet Answers
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Bond Gizmo Worksheet Answers
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Covalent Bond Gizmo Worksheet Answers
- Setting Reading Goals Covalent Bond Gizmo Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Covalent Bond Gizmo Worksheet Answers
- Fact-Checking eBook Content of Covalent Bond Gizmo Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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