## Covalent Bonding Worksheet

Covalent bonding occurs when two or more NON METALS share electrons, attempting to attain a stable octet (8 outer electrons)in their outer shell for at least part of the time. Draw a Lewis dot diagram for each element listed. Circle the unpaired electrons that will be shared between the elements.

H<sub>2 bydrogen</sub> is distornic

H + H → H → H-H Single Bond

F<sub>2</sub> theories in distance.

3.) O2 oxygen is diatomic

O + O

\* O=O

Double Bond

4.) No nitrogen is dianomic. In this a triple bond?

N + N

5.) BF<sub>3</sub> you need 3 fluorine atoms here

 $\mathbf{R} + \mathbf{F}$ 

6.) Ammonia NH3 hint: how many hydrogen atoms are needed?

N + H

7.) Carbon dioxide CO-

0 + c + 0

Methane CH<sub>4</sub> careful here 4 hydrogen atoms needed

C + H

9.) Dihydrogen monoxide: the most dangerous substance on the planet. It has killed more people than any other substance known to mankind!!

H + O + H

10.) SO2 hint: one pair of electrons from sulfur must be slit up for this one to work.

O + S + O

# **Covalent Bonding Assessment Answers**

**Hussin A.Rothana** 

#### **Covalent Bonding Assessment Answers:**

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