

ACTIVITY SERIES WORKSHEET ANSWERS

Complete the table below using the activity series of metals

	Na	Zn	Al	Cu
CuCl_2	$\text{Na} + \text{CuCl}_2$ \rightarrow $\text{NaCl}_2 + \text{Cu}$	$\text{Zn} + \text{CuCl}_2$ \rightarrow $\text{ZnCl}_2 + \text{Cu}$	$\text{Al} + \text{CuCl}_2$ \rightarrow $\text{AlCl}_3 + \text{Cu}$	NR
$\text{Fe}(\text{NO}_3)_3$	$\text{Na} + \text{Fe}(\text{NO}_3)_3$ \rightarrow $\text{NaNO}_3 + \text{Fe}$	$\text{Zn} + \text{Fe}(\text{NO}_3)_3$ \rightarrow $\text{Zn}(\text{NO}_3)_2 + \text{Fe}$	$\text{Al} + \text{Fe}(\text{NO}_3)_3$ \rightarrow $\text{Al}(\text{NO}_3)_3 + \text{Fe}$	NR
AgNO_3	$\text{Na} + \text{AgNO}_3$ \rightarrow $\text{NaNO}_3 + \text{Ag}$	$\text{Zn} + \text{AgNO}_3$ \rightarrow $\text{Zn}(\text{NO}_3)_2 + \text{Ag}$	$\text{Al} + \text{AgNO}_3$ \rightarrow $\text{Al}(\text{NO}_3)_3 + \text{Ag}$	NR
H_2SO_4	$\text{Na} + \text{H}_2\text{SO}_4$ \rightarrow $\text{Na}_2\text{SO}_4 + \text{H}_2$	$\text{Zn} + \text{H}_2\text{SO}_4$ \rightarrow $\text{ZnSO}_4 + \text{H}_2$	$\text{Al} + \text{H}_2\text{SO}_4$ \rightarrow $\text{Al}_2(\text{SO}_4)_3 + \text{H}_2$	NR
$\text{Al}(\text{OH})_3$	$\text{Na} + \text{Al}(\text{OH})_3$ \rightarrow $\text{NaOH} + \text{Al}$	NR	NR	NR

Applying The Activity Series Worksheet Answers

Paul Bramley



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