Student Exploration: Limiting Reactants

Vocabulary: chemical equation, chemical formula, chemical reaction, coefficient, limitingreactant, molecule, product, reactant, subscript

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)
Imagine you and your friends are making hot dogs. A complete hot dog consists of a wiener and a bun. At the store, you buy four packages of eight wieners and three bags of 10 buns.

- How many total hot dogs can you make? _____you can make 30 hotdogs with single bun and wieners _____
- Which ingredient limited the number of hot dogs you could make? the buns having 30 limited the number of hot dog i can make _____
- Which ingredient will you have leftovers of? i would have leftover of the wieners remaining 2 more.

Gizmo Warm-up

Just as ingredients can be put together to make a new food, substances can combine during a **chemical reaction** to produce new substances. The substances that undergo change are called **reactants**. The new substances are **products**.

Sometimes during a chemical reaction, one type of reactant will be used up before the other reactants. This reactant is the **limiting reactant**. Using the *Limiting Reactants* Gizmo, you can determine which reactant is limiting in various scenarios.



To begin, make sure $H_2 + O_2$ becomes H2O. The small "2" in H_2 , O_2 , and H_2O is a subscript. Subscripts represent the number of atoms in a **molecule**.

- Use the sliders to set the number of O₂ molecules and H₂ molecules to two.
 - A. How many hydrogen molecules (H₂) are there?____1_Hydrogen atoms?
 - B. How many oxygen molecules (O2) are there?___1__Oxygen atoms?___2__
- How many H₂O molecules do you think will form when these four molecules react?



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