

TED-Ed: The Strengths and Weaknesses of Acids and Bases

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1. Before beginning the video, what do you already know about acids and bases? Explain your knowledge base.

a. What are some items that acids and bases are used to make? List them.

b. A long time ago, before anyone knew about atoms or molecules, acids and bases were defined by how they _____.

c. When molecules dissolved in water interact, they're exchanging two main currencies with their surroundings: _____ (also known as hydrogen ions) and _____.

i. True or false (circle one): Depending on how a molecule is composed or shaped, it may be willing to donate or accept either protons or electrons with some other community member.

ii. Some substances are so aggressive about donating their protons that when they get a chance, all of the molecules in a sample will dump a proton, sometimes more than one, to the surrounding water molecules; we call these _____.

1. Comparatively, what are strong bases? Explain.

2. What are the characteristics of weak acids and weak bases?

a. If left alone in water, they reach some _____ point where maybe only one out of a hundred or one out of ten thousand of their molecules has exchanged currency with water.

d. Most acid-base chemistry takes place in _____ which can act as either an acid or a base.

i. When net effect on water's account cancels out, we call this _____.



193 Strengths Of Acids And Bases Worksheet

Ronald Percy Bell, R. P. Bell



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