

## Practice Problems

Do each problem using the following guidelines:

## SECTION 15.1 ACID-BASE THEORIES

- Classify the acid-base reaction according to Lewis theory as acid-base, Lewis acid-base, Lewis conjugate acid-base, or Lewis acid-base conjugate acid-base.
- Classify it in the reaction that it does *not* participate in.
- Classify the reaction in the appropriate Lewis conjugate acid-base reaction.
  - Acid: \_\_\_\_\_ Base: \_\_\_\_\_
  - Acid: \_\_\_\_\_ Base: \_\_\_\_\_
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{HCl}$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{NaOH}$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{H}_2\text{SO}_4$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{NaHCO}_3$  in water.

## SECTION 15.2 HYDROGEN ION AND ACIDITY

- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{HCl}$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{NaOH}$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{H}_2\text{SO}_4$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{Na}_2\text{CO}_3$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{NaHCO}_3$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{H}_2\text{PO}_4^-$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{HPO}_4^{2-}$  in water.
- Calculate the  $\text{pH}$  of a 0.1 M solution of  $\text{H}_2\text{PO}_4^-$  in water.
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# 19 Acids Bases And Salts Practice Problems

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