

Name: _____

Date: _____

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. There are eight markers in a full set, but Flora and Frank each only have seven markers. Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? **They can share the markers.**

2. Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

They can give the ones they need to each other.

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the *Covalent Bonds* Gizmo, atoms form bonds in this way.

To begin, check that **Fluorine** is selected from the **Select a substance** menu. Click **Play** () to see the electrons orbiting the nucleus of each atom.



1. The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have? _____7_____
2. Click **Pause** (). Drag an electron from the left atom to the right atom. Click **Play**.
What happens? **They share an electron.**
3. Click **Pause**, drag an electron from the right atom to the left, and then click **Play**.
What happens now? **They share another one.**

Covalent Bond Gizmo Worksheet Answer Key

BM King



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Table of Contents Covalent Bond Gizmo Worksheet Answer Key

1. Understanding the eBook Covalent Bond Gizmo Worksheet Answer Key
 - The Rise of Digital Reading Covalent Bond Gizmo Worksheet Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Covalent Bond Gizmo Worksheet Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Covalent Bond Gizmo Worksheet Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Covalent Bond Gizmo Worksheet Answer Key
 - Personalized Recommendations
 - Covalent Bond Gizmo Worksheet Answer Key User Reviews and Ratings
 - Covalent Bond Gizmo Worksheet Answer Key and Bestseller Lists
5. Accessing Covalent Bond Gizmo Worksheet Answer Key Free and Paid eBooks
 - Covalent Bond Gizmo Worksheet Answer Key Public Domain eBooks
 - Covalent Bond Gizmo Worksheet Answer Key eBook Subscription Services
 - Covalent Bond Gizmo Worksheet Answer Key Budget-Friendly Options

6. Navigating Covalent Bond Gizmo Worksheet Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Covalent Bond Gizmo Worksheet Answer Key Compatibility with Devices
 - Covalent Bond Gizmo Worksheet Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Covalent Bond Gizmo Worksheet Answer Key
 - Highlighting and Note-Taking Covalent Bond Gizmo Worksheet Answer Key
 - Interactive Elements Covalent Bond Gizmo Worksheet Answer Key
8. Staying Engaged with Covalent Bond Gizmo Worksheet Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Covalent Bond Gizmo Worksheet Answer Key
9. Balancing eBooks and Physical Books Covalent Bond Gizmo Worksheet Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Covalent Bond Gizmo Worksheet Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Covalent Bond Gizmo Worksheet Answer Key
 - Setting Reading Goals Covalent Bond Gizmo Worksheet Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Covalent Bond Gizmo Worksheet Answer Key
 - Fact-Checking eBook Content of Covalent Bond Gizmo Worksheet Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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