Unit 1: Matter and Energy

Name: KEY
Date: Class:

<u>Directions</u>: Complete the following review questions. Review your Cornell notes, learning targets, worksheets, labs, and other assignments to adequately prepare for the unit test.

Learning Target(s): 1, 2, 3, 4, 10

- In the Melting of a Solid lab, you heated (or cooled) an unknown solid and recorded its temperature at 30-second intervals. From this data you created a heating (or cooling) curve to identify the substance.
 - a. Identify the independent variable of this experiment: the time you heat the solid for
 - b. Identify the dependent variable of this experiment: the temperature of the solid
 - c. Identify the control for this experiment; substance used; method of heating
- 2. Decide whether the following observations are qualitative or quantitative:
 - The solution began to get cloudy once the second reagent was added. Qualitative
 - The piece of magnesium was 1.25 centimeters in length. Quantitative
 - The mass of the block decreased by fifteen grams. Quantitative
 - d. The test tube appeared to have a dark orange solid suspended in the solution. Qualitative
 - e. The residue left in test tube 5 was slightly larger than in test tube 4. Qualitative

Learning Torget(s): 15

- 3. Suggest a process to use to separate each of the following mixtures:
 - a. A mix of salt and water:
 - b. A smoothie drink with veggies and fruits of different densities:
- 4. Which physical property does filtration use to separate a mixture?

Learning Target(s): 5, 6

- 5. Describe at least two safety precautions to use when using a Bunsen Burner in lab:
 - a. Tie back hair and loose clothing
 - b. Light match first then turn on gas valve
- Which type of lab equipment would be most appropriate to measure 4 mL of water in? Choose one and explain why.
 - n. A 250 mL beaker

d. A 50 mL graduated cylinder

b. A 10 mL graduated cylinder

e. A large test tube

c. An electronic balance

f. A thermometer

You would get the most accurate measurement with this size. This piece of equipment has graduations to measure volume.

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Marcel A. Müller

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Table of Contents Science Review Sheet Unit 1 Freshman

- 1. Understanding the eBook Science Review Sheet Unit 1 Freshman
 - The Rise of Digital Reading Science Review Sheet Unit 1 Freshman
 - o Advantages of eBooks Over Traditional Books
- 2. Identifying Science Review Sheet Unit 1 Freshman
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Science Review Sheet Unit 1 Freshman
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Science Review Sheet Unit 1 Freshman

- Personalized Recommendations
- Science Review Sheet Unit 1 Freshman User Reviews and Ratings
- Science Review Sheet Unit 1 Freshman and Bestseller Lists
- 5. Accessing Science Review Sheet Unit 1 Freshman Free and Paid eBooks
 - Science Review Sheet Unit 1 Freshman Public Domain eBooks
 - Science Review Sheet Unit 1 Freshman eBook Subscription Services
 - Science Review Sheet Unit 1 Freshman Budget-Friendly Options
- 6. Navigating Science Review Sheet Unit 1 Freshman eBook Formats
 - o ePub, PDF, MOBI, and More
 - Science Review Sheet Unit 1 Freshman Compatibility with Devices
 - Science Review Sheet Unit 1 Freshman Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Science Review Sheet Unit 1 Freshman
 - Highlighting and Note-Taking Science Review Sheet Unit 1 Freshman
 - Interactive Elements Science Review Sheet Unit 1 Freshman
- 8. Staying Engaged with Science Review Sheet Unit 1 Freshman
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Science Review Sheet Unit 1 Freshman
- 9. Balancing eBooks and Physical Books Science Review Sheet Unit 1 Freshman
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Science Review Sheet Unit 1 Freshman
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Science Review Sheet Unit 1 Freshman
 - Setting Reading Goals Science Review Sheet Unit 1 Freshman
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Science Review Sheet Unit 1 Freshman

- Fact-Checking eBook Content of Science Review Sheet Unit 1 Freshman
- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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