CHAPTER - 2
STRUCTURE OF ATOM
Discovery of electrion (Cathode Rays)
Cathode stays > William (stookes (1879) Electron (ie) > J. J. Thomson, (-ve charge particle)
Electron Gery J. J. Thomson,
in discharge trube known as cathode stays.
in discharge trube known as cathode grays.
(m) at Ind Port To Vaccion pump
Coul at low love.
Constant of the post of the pump
homograpical High Wolfbor
properties righ voltage
(i) Calledo seems domical in a stanight line (shadow)
(2) Cathode graye carry -ve charge consistive field all the
(3) Cathode 21998 and deflected by majorite field
(4) They are made up of material particle (Potato 1900)
(5) They produce heating effect & possess penetorating effect
(6) They produce green flyonescence & X- mays
(5) They produce heating effect & possess peneterating effect. (6) They produce green flyoriescence & x- rays. (7) They also affect photographic plate.
Discovery of prioton (Anode / Canal Rays)
Annde marie -> (moldstein (1886)
Anode rays -> Goldstein (1886) Proton (18) -> Goldstein & named by Rytherford (+ve change fotich)
the charged particle which more opposite the cathode
the charged particle which move opposite the cathode in discharge tube known as andde stays.

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JR Anderson

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