

AP Chemistry Practice Test Chapter 8 and 9

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- ____ 1. Which pair of atoms should form the most polar bond?
a. F and B
b. C and O
c. F and O
d. N and F
e. B and N
- ____ 2. Which pair of ions should form the ionic lattice with the highest energy?
a. Na^{1+} and Br^{1-}
b. Li^{1+} and F^{1-}
c. Cs^{1+} and F^{1-}
d. Li^{1+} and O^{2-}
e. K^{1+} and F^{1-}
- ____ 3. The electrostatic force of attraction is greatest in which compound?
a. BaO
b. MgO
c. CaS
d. MgS
e. CaO
- ____ 4. Which molecule has the weakest bond?
a. CO
b. O_2
c. NO
d. N_2
e. Cl_2
- ____ 5. For which species are octet resonance structures necessary and sufficient to describe the bonding satisfactorily?
a. BCl_3
b. SO_2
c. CO_2
d. BeF_2
e. SO_4^{2-}
- ____ 6. How are the bonding pairs arranged in the best Lewis structure for ozone?
a. O—O—O
b. O=O—O
c. O=O—O
d. O=O=O
e. O=O=O
- ____ 7. Which species has the shortest bond length?
a. CN^{1-}
b. O_2
c. SO_2
d. SO_3
e. CO_2
- ____ 8. Which species requires the least amount of energy to remove an electron from the outermost energy level?
a. Na^{1+}
b. He
c. F^{1-}
d. O^{2-}
e. Mg^{2+}
- ____ 9. Which species has a valid non-octet Lewis structure?
a. GeCl_4
b. SiF_4
c. CO_3^{2-}
d. SeCl_4
e. CO_3^{2-}

Chemistry Practice Test A Pearson Answers

JoAnn Zerwekh



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Master the PCAT Peterson's, 2012-07-15 Peterson's *Master the PCAT* is an in depth review that offers thorough preparation for the computer based exam After learning about the structure format scoring and score reporting and the subtests and question types you can take a diagnostic test to learn about your strengths and weaknesses The next six parts of the eBook are focused on detailed subject reviews for each subtest verbal ability reading comprehension biology chemistry quantitative ability and writing Each review includes practice questions with detailed answer explanations You can take two practice tests to track your study progress The tests also offer detailed answer explanations to further improve your knowledge and understanding of the tested subjects The eBook concludes with an appendix that provides helpful information on a variety of careers in pharmacy and ten in depth career profiles **A Crash Course in AIEEE Chemistry 2011**, A Crash Course In Aieee Chemistry 2009 Arora, 2009-09 *Chemistry: The Central Science* Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Dalius Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual understanding development of problem solving skills reference and test preparation

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