

Chemistry Unit 8 Worksheets & Examples of Every Kind of Problem

- 1.** **Equation:** $20\text{AgNO}_3 + 10\text{AgCl} \rightarrow 30\text{Ag} + 20\text{Cl}^-$

Reactants: 20.000 g 10.000 g 30.000 g

Products: 20.000 g 10.000 g 30.000 g

Mass: 30.000 g 30.000 g 30.000 g

0.20 moles used 0.10 moles generated
- 2.** **Equation:** $2\text{Ag} + 2\text{HCl} \rightarrow 2\text{AgCl} + \text{H}_2$

Reactants: 10.000 g 10.000 g 10.000 g

Products: 10.000 g 10.000 g 10.000 g

0.200 moles used 0.100 moles generated
- 3.** **Equation:** $2\text{AgNO}_3 + 2\text{HCl} \rightarrow 2\text{AgCl} + 2\text{HNO}_3$

Reactants: 20.000 g 10.000 g 30.000 g

Products: 20.000 g 10.000 g 30.000 g

0.10 moles used 0.10 moles generated
- 4.** **Equation:** $2\text{AgNO}_3 + 2\text{HCl} \rightarrow 2\text{AgCl} + 2\text{HNO}_3$

Reactants: 10.000 g 10.000 g 20.000 g

Products: 10.000 g 10.000 g 20.000 g

0.10 moles used 0.10 moles generated
- 5.** **Equation:** $2\text{AgNO}_3 + 2\text{HCl} \rightarrow 2\text{AgCl} + 2\text{HNO}_3$

Reactants: 10.000 g 10.000 g 20.000 g

Products: 10.000 g 10.000 g 20.000 g

0.10 moles used 0.10 moles generated

Chemistry Unit 8 Ws 2 Answers

Career Point Kota



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Maths in Chemistry Prerna Bansal, 2024-09-02 Numerical methods are the mathematical procedures that approximate the solution of complex mathematical problems into much simpler form and which find a wide variety of use while solving complex Physical Chemistry problems This book aims to aide in understanding of such numerical methods including solving complex differential equations and numerical differentiation integration Moreover it also explains various statistical tests used in Analytical Chemistry for data analysis The author has tried to include as many example from Chemistry problems for a better understanding of the methods

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